

All-Optical Emission Control and Lasing in Plasmonic Lattices

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Abstract

We report on reversible all-optical emission control and lasing in plasmonic nanoparticle lattices. By incorporating photochromic molecules into the liquid gain medium composed of organic fluorescent molecules, we realize all-optical control over gain and absorption, the two key parameters associated with both conventional and nanoscale lasing. We demonstrate reversible photoswitching between two distinct modes of operation, 1) spontaneous emission to the lattice mode, characterized by broad emission linewidth, low emission intensity and large angular distribution and 2) lasing action, characterised by very narrow (sub-nm) linewidths due to emergence of increased gain and temporal coherence in the system, approximately three orders of magnitude increase in emission intensity, and narrow 0.7 degree angular divergence of the beam. A rate-equation model is employed to describe the operation of the switchable plasmonic laser. Our results provide the first demonstration of optically tunable losses

in plasmonic lattice lasers, which is an important milestone for the development of active plasmonics and paves the way for ultrafast all-optical switching of plasmonic nanolasers.

Keywords

Plasmonics, nanolasing, photoswitching, surface lattice resonance, optical control

Introduction

Metallic nanoparticles support surface plasmon resonances (SPRs) that exhibit strong near-field character and tunability of the resonance condition *via* material choice, particle geometry, and refractive index of the surroundings. This makes SPRs a promising platform for nanoscale sensing applications.^{1,2} However, the high radiative and ohmic losses of SPRs have significantly limited the range of applicability of these resonances. In metallic nanoparticle arrays, the SPR of an individual particle can couple to other particles *via* radiation fields. The radiation-induced coupling of the lossy SPR mode with the low-loss diffracted order of the lattice induces hybrid modes known as surface lattice resonances (SLRs), with significantly reduced linewidths³⁻⁸ as compared to their uncoupled counterparts. Further, the tunability of the band structure *via* particle and lattice geometry provides a versatile platform for studying light-matter interactions at the nanoscale.⁹

Lasing action in IR and visible has been reported in nanoparticle arrays overlaid with optically pumped fluorescent molecules. Linewidth narrowing, increased temporal and spatial coherence, highly directional beam, and several orders of magnitude increase in emission intensity (for a given wavelength) was observed upon the onset of lasing.^{10,11} This together with the previous observations of lasing in plasmonic systems,^{12,13} provided an important milestone towards applications where plasmonic losses have typically been considered a hin-

drance. Recently, plasmonic nanoparticle arrays have been shown to exhibit wavelength-tunable lasing when varying the refractive index of the gain medium,¹⁴ or the interparticle spacing on a stretchable substrate.¹⁵

Controlling lasing optically is desirable for the design of novel devices such as optical switches and memories.¹⁶ All-optical switching of lasing has been referred to modifying either the output intensity characteristics or the direction of lasing operation.¹⁷⁻¹⁹ In principle, tuning optical properties of lasing other than amplitude or directionality (such as phase, polarization, spatial/temporal distribution, *etc.*) could be beneficial for applications. In this work we focus on the on-off switching of lasing with photochromic molecules.

Photochromic molecules change their conformation reversibly upon exposure to light, modifying their absorption and emission spectra.²⁰ Photochromic molecules have been utilized, for instance, in optical switches^{21,22} and transistors.^{23,24} They are also used for controlling the strength of light-matter interaction.^{25,26} Controlling optical gain by photochromic molecules was first proposed in a system consisting of photochromic molecules in a solid-state polymeric matrix.¹⁶ In the experiment, Pisignano *et al.* showed how amplified spontaneous emission can be switched on and off by all-optical control. Photoswitching of actual lasing was first reported in photonic crystals.^{27,28} Optical amplification at low temperatures²⁹ and at room temperature³⁰ has been demonstrated in optical microcavities. Very recently, even a single photon induced switching of lasing has been observed.³¹ Finally, a microcavity-based serial combination of all-optical components has been demonstrated,³² a necessary prerequisite for more complex logical operations. Plasmonic photoswitchable lasing has been theoretically proposed for multilayer plasmonic-photonic structures,³³ and a photoswitchable spaser was experimentally demonstrated with gold nanoparticles in polymer shell covered with fluorescent proteins.³⁴

Here, we demonstrate all-optical switching of lasing in plasmonic nanoparticle lattices using photochromic spiropyran derivatives. Unlike in Ref.³⁴ where the wavelength of lasing was optically controlled, we demonstrate on-off switching of lasing in a plasmonic lattice.

Our system is employing simple, commercially available compounds together with collective SLR modes instead of single particle plasmon resonances, allowing for a strong modulation of the beam divergence between the two regimes of operation. The high intensity modulation enabled by the system may be relevant for developing ultra-fast nanoscale optical circuits analogous to electronic binary circuits.³⁵

Our sample consists of a square array of Ag nanoparticles (Fig. 1 (a)) overlaid with a liquid medium consisting of organic coumarin 6 (C6) molecules providing the optical gain, and photochromic indolinospiropyran molecules, which, depending on their conformational state, can be either strongly absorbing or nearly transparent at the lasing wavelength. The state can be reversibly changed by exposure to UV and visible light, see Fig. 1 (b), providing all-optical control of the absorption and emission properties of the hybrid system. A schematic of the sample is shown in Fig. 1 (c).

Results

When the particle periodicity p in a plasmonic lattice equals the wavelength of radiation λ in a medium with refractive index n , the radiation fields from all the particles interfere constructively at each particle location, creating a constant phase and increased dipole moment in each particle at $k = 0$ at a wavelength $\lambda = p \times n$, *i.e.* at the Γ -point. This radiative coupling of individual nanoparticles can be used as a feedback mechanism for lasing and condensation.^{11,36-39}

The absorbance of the photochromic molecule, indolinospiropyran (1',3'-dihydro-1',3',3'-trimethyl-6-nitrospiro [2H-1-benzopyran-2,2'-(2H)-indole]), depends strongly on the conformational state of the molecule. In its relaxed closed form, spiropyran (SPI) is transparent across the visible light spectrum, whereas the open form merocyanine (MC) absorbs strongly at around 2.28 eV. The switching from pure SPI to a photostationary state containing *ca.* 14% MC (see Section S1 in the Supporting Information) is performed by UV light (365 nm

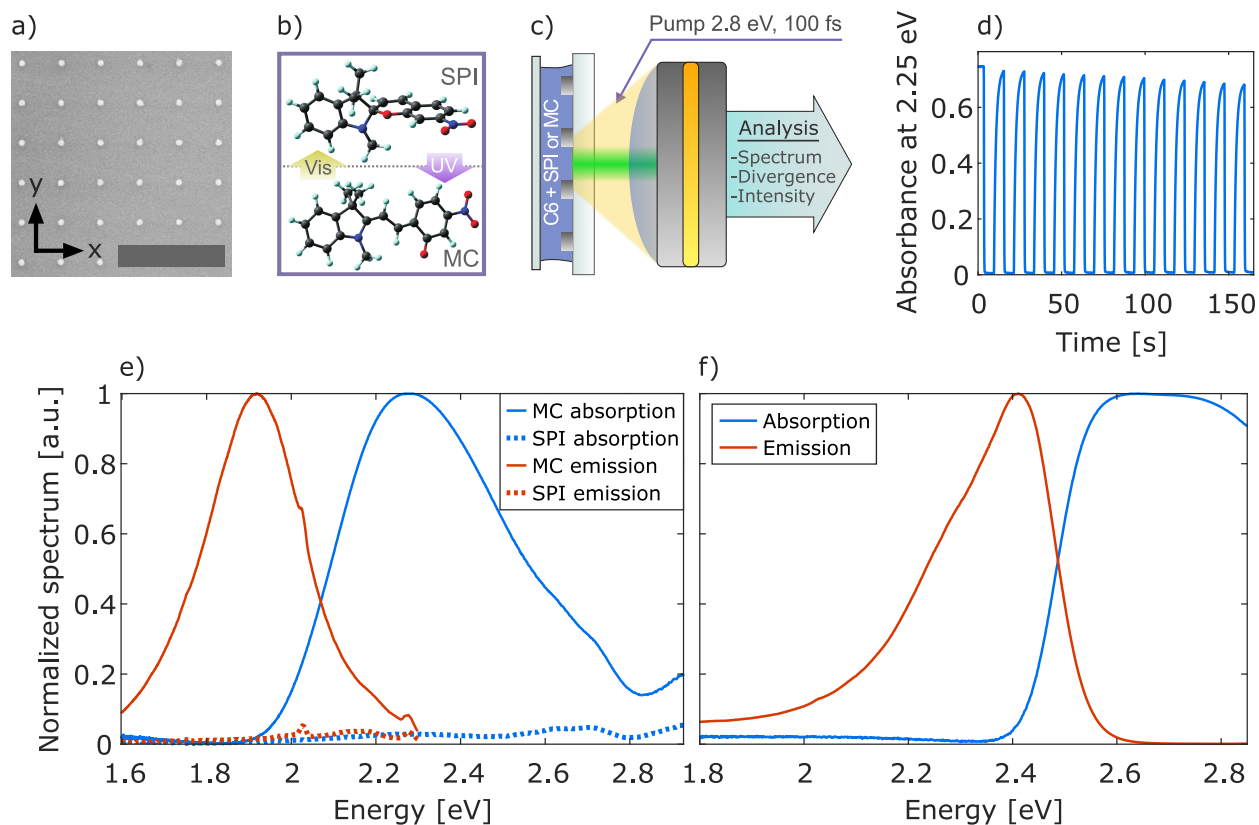


Figure 1: (a) A scanning electron micrograph of a typical sample, scale bar 1 μm . See Methods for description of the sample fabrication. (b) Illumination-induced isomerizations between SPI and MC. (c) For lasing measurements, a nanoparticle lattice overlaid with gain solution was pumped optically and the resulting emission was collected for angle and wavelength resolved data analysis. See Methods and Section S5 in the Supporting Information for details of the experimental setup. (d) Reversible absorbance of the photochromic molecules at 2.25 eV during alternating UV and visible light treatments. See Methods for details of the spectrophotometric measurement. (e) The absorption (blue) and emission (red) of the pure SPI isomer (dashed) and the photostationary mixture of SPI and MC (solid). The used concentration in (d-e) was the same as in lasing measurements (40 mM). (f) The absorption (blue) and emission (red) of C6 (30 mM). The used solvent in (d-f) was benzyl alcohol.

wavelength, 50 W/m^2), and reversed by visible (green) light illumination. Fig. 1 (d) shows the reversible absorbance of the photochromic molecule during consecutive UV and visible light treatments. The MC–SPI isomerization also occurs *via* thermal relaxation but at room temperature it is significantly slower compared to using visible light illumination (see Section S2 in the Supporting Information). The normalized emission and absorption profiles of the SPI and MC forms are shown in Fig. 1 (e) which highlights the drastic difference between the two forms in terms of optical properties in the visible spectrum.

Fig. 1 (f) shows the normalized emission and absorption profiles of the used gain material C6, which was chosen as its emission profile overlaps with the absorption profile of MC. Additionally, the absorption profile of C6 allows it to be pumped at 450 nm ($\approx 2.8 \text{ eV}$), which minimizes the effect of the pump laser on the form of the photochromic molecules as the absorbance of both SPI and MC is close to zero at this wavelength. Although the emission from C6 can cause the isomerization of some MC molecules back to the SPI form, the continuous replenishment of molecules from the dye bath prevents this from significantly changing the system operation in the timescale of our measurements. The plasmonic lattice optimized for lasing has a lattice periodicity of 350 nm which sets the Γ -point energy at 2.3 eV with a refractive index of 1.54 in the fluorescent medium. The array size is $100 \times 100 \mu\text{m}^2$, and the cylindrical nanoparticles have a diameter and height of 60 nm and 30 nm, respectively.

The composite gain/loss medium consisting of C6 and photochromic molecules (in 30 mM : 40 mM ratio), was separately characterized, see Fig. 2. A small volume (7 μl) of the mixture was injected between a borosilicate microscope slide and a $22 \times 22 \text{ mm}^2$ cover slip, and the emission was measured with several different UV exposure times, see Fig. 2 (a). In the absence of UV exposure, the emission resembles that of the C6 molecules (red dashed curve) with the maximum at around 2.41 eV. Increasing UV exposure gradually decreases the emission intensity, which is associated with the increased absorbance of the photochromic molecules. With sufficient (30 s) UV exposure (red solid curve), the emission reaches a minimum with approximately 20% of the initial emission intensity. Further, emission at 1.9 eV

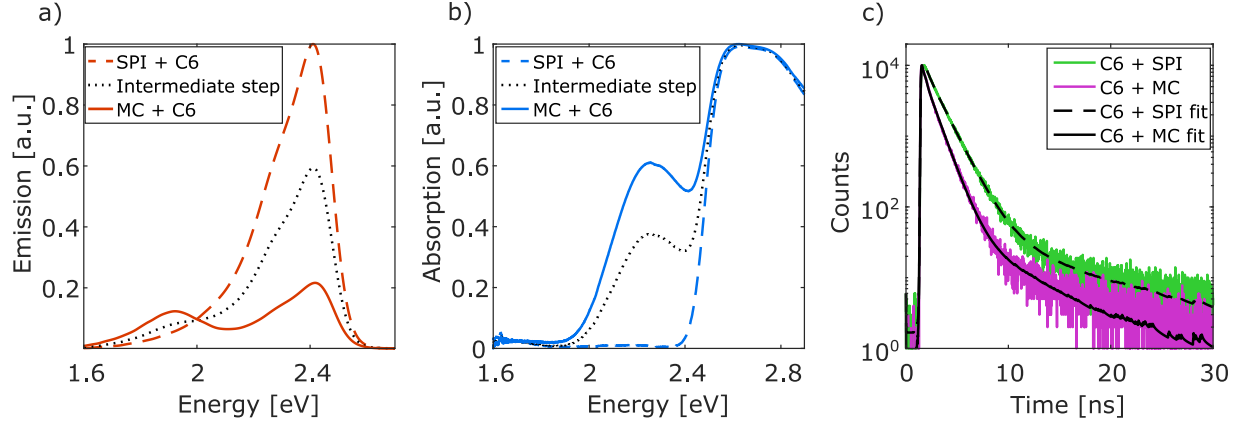


Figure 2: Emission and absorption properties of the composite gain medium. (a) Emission profile before (red dashed line) and after (red solid line) UV treatment. (b) Absorption before (blue dashed line) and after (blue solid line) UV treatment. The black dashed line in (a) and (b) shows an intermediate step between SPI and MC forms. (c) Lifetime measurements of the composite gain medium with the photochromic molecules in the SPI (green line) and MC (purple line) forms. Multiexponential fits to the lifetime data in SPI and MC forms are shown as dashed and solid black lines, respectively. The oscillations visible in the fitting curves above 20 ns are caused by background noise in the instrumental response function. See Methods for details of the time-resolved fluorescence spectroscopy setup.

is observed which we associate with the fluorescence of the MC form of the photochromic molecule. The absorption profile with similar configurations is shown in Fig. 2 (b). Before UV exposure (dashed blue line), the absorption resembles that of C6 but after sufficient UV treatment (blue solid line) the absorption peak of MC appears. In contrast to the measurement shown in Fig. 1 (d), here we use significantly longer exposure times for molecule switching as we wish to ensure maximal isomerization of the photochromic molecules. Due to the high concentration of both C6 and photochromic molecules in the solution, there is a possibility for non-radiative energy transfer between them. This was confirmed with time-resolved fluorescence spectroscopy: the fluorescence lifetime of C6 decreased from 1.42 to 1.06 ns upon SPI-MC isomerization (see Fig. 2 (c) and Section S3 in the Supporting Information). We note that the exact mechanism of the energy transfer requires further studies, and that both Förster- and Dexter-type mechanisms are plausible under these conditions.

For the lasing measurements, we overlay the nanoparticle array with the composite fluorescent solution and optically pump the structure with a femtosecond laser (450 nm center

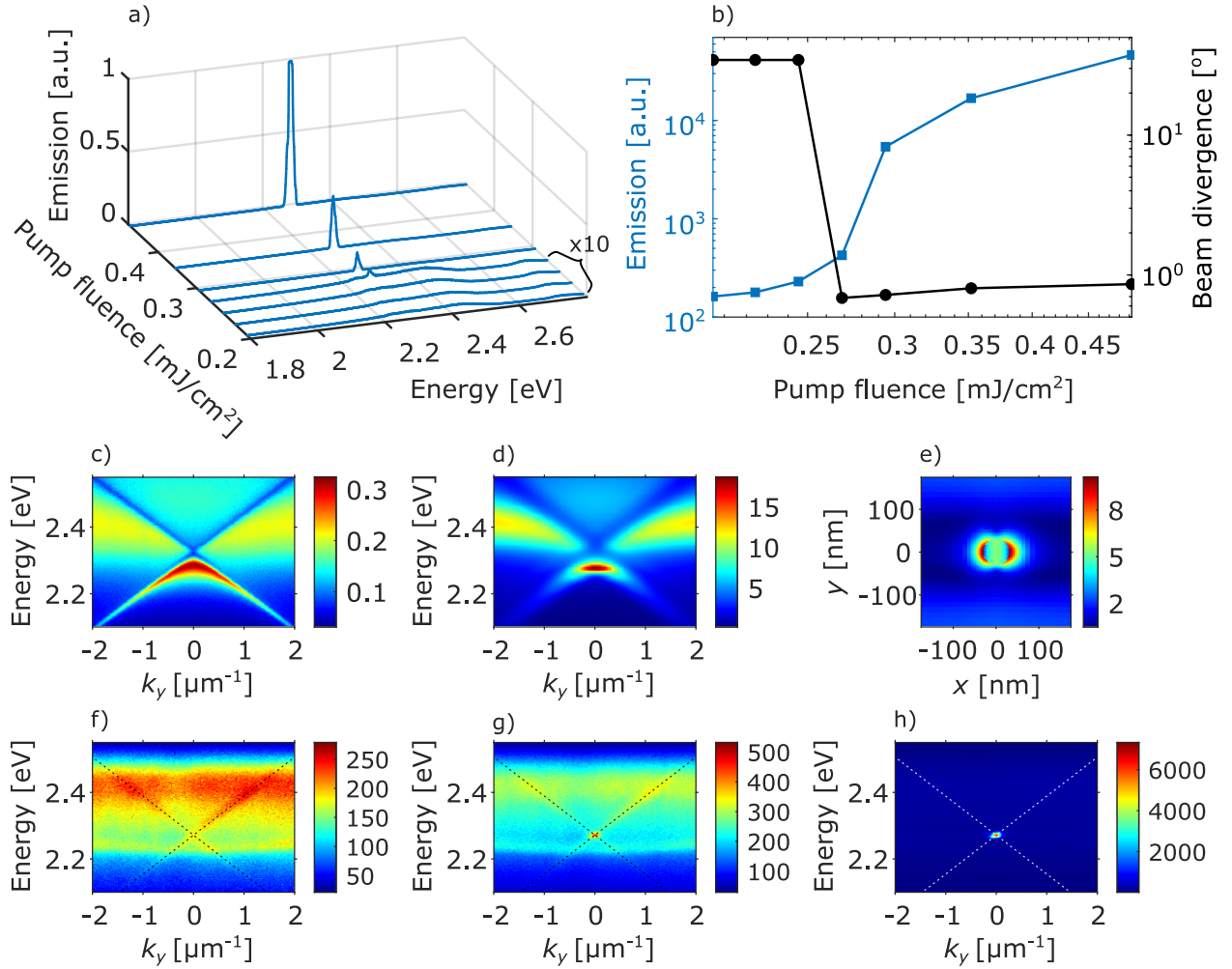


Figure 3: Emission below and above the lasing threshold. (a) Emission spectra as a function of energy for different pump fluences. (b) Integrated emission intensity (blue squares) near the Γ -point and beam divergence (black circles) at $k_y = 0$ as a function of pump fluence. (c) Measured extinction of a typical bare array without fluorescent molecules calculated as $(1 - T)$, where T is transmission. (d) Simulated extinction efficiency of the array. The array period is 350 nm and the particle radius and height are 30 nm. The simulation was done by using the coupled dipole approximation (CDA),⁴⁰ see Methods. (e) The finite-difference time-domain (FDTD) simulation of a nanoparticle array illuminated with x -polarized plane wave at 2.3 eV. The color scale corresponds to the electric field profile $|E|$ normalized by the amplitude of the incoming plane wave. See Methods for details of the simulation. Also shown are the measured angle and energy resolved intensity data (f) below, (g) at and (h) above threshold pump fluence.

wavelength, 100 fs pulse duration, 1 kHz repetition rate). In Fig. 3 (a) the sample emission is shown as functions of energy and pump fluence, exhibiting a transition between two regimes: at low pump fluences, the emission intensity is low and the spectra are broad due to spontaneous emission. At higher pump fluences, though, the emission intensity increases in a nonlinear fashion, which is also evident from Fig. 3 (b). We note that this transition is accompanied with a rapid reduction in the emission beam divergence down to 0.7° . For the emission linewidths, see Section S6 in the Supporting Information.

Experimentally obtained and simulated dispersions of the array optimized for lasing are shown in Figs. 3 (c-d). The lattice dispersion shows the crossing of the $\langle 0, +1 \rangle$ and $\langle 0, -1 \rangle$ diffracted orders of the lattice at the Γ -point. Fig. 3 (e) shows the finite-difference time domain (FDTD) simulation of the nanoparticle array illuminated with a plane wave at the Γ -point energy. In Fig. 3 (f-h), we plot the angle and wavelength resolved emission data, revealing the crucial role of the nanoparticle lattice in both spontaneous emission and lasing regimes. The spontaneous emission angles are modified by the presence of the lattice, with the maximum emission taking place at the angles and energies that correspond to the SLR mode (dashed lines in Fig. 3 (f-h)). In the lasing regime, vast majority of photons are occupying a narrow energy ($\Delta E = 11$ meV) and wavevector ($\Delta k_y = 0.137 \mu\text{m}^{-1}$) range at the Γ -point of the nanoparticle lattice.

In order to simulate the composite system response, we use a rate equation model consisting of a 4-level system as shown in Fig. 4 (a). The model is defined with the following coupled rate equations:

$$\frac{dn_{\text{ph}}}{dt} = \beta(n_{\text{ph}} + 1)\frac{N_2}{\tau_{21}} - \beta n_{\text{ph}}\frac{N_1}{\tau_{21}} - \frac{n_{\text{ph}}}{\tau_{\text{ph}}} \quad (1)$$

$$\frac{dN_0}{dt} = -P(t)N_0 + \frac{N_2}{\tau_{20}} + \frac{N_1}{\tau_{10}} \quad (2)$$

$$\frac{dN_1}{dt} = \beta n_{\text{ph}}\frac{(N_2 - N_1)}{\tau_{21}} + \frac{N_2}{\tau_{21}} - \frac{N_1}{\tau_{10}} \quad (3)$$

$$\frac{dN_2}{dt} = -\beta n_{\text{ph}}\frac{(N_2 - N_1)}{\tau_{21}} - \frac{N_2}{\tau_{21}} - \frac{N_2}{\tau_{20}} + \frac{N_3}{\tau_{32}} \quad (4)$$

$$\frac{dN_3}{dt} = P(t)N_0 - \frac{N_3}{\tau_{32}}, \quad (5)$$

where the populations of each level are denoted with N_i and the lifetimes of transitions between levels i - j with τ_{ij} . Here, n_{ph} is the photon number in the lasing mode. Pump rate $P(t)$ is proportional to the pump intensity that has a Gaussian temporal profile centered at 150 fs and full width at half maximum of 100 fs. The lifetime of the SLR mode $\tau_{\text{ph}} = 85$ fs is obtained from the dispersion linewidth at $k_y = 0$ in Fig. 3 (c). The spontaneous emission factor β is given by the ratio between spontaneously emitted photons and all photons emitted into the lasing mode.^{41,42} Increase of total intensity of about 2 orders of magnitude in Fig. 3 (b) corresponds to $\beta = 0.01$.

The 2-1 transition of the 4-level system is resonant with the cavity mode and therefore corresponds to the radiative spontaneous emission lifetime of C6 (τ_{21}). The 2-0 transition is non-resonant and relates to a non-radiative lifetime (τ_{20}). Intrinsic non-radiative decay of C6 is on the order of 11.8 ns⁴³ which is significantly longer than any other time scale in the system dynamics. Therefore, we neglect the effect of the intrinsic non-radiative decay and use the parameter τ_{20} to model the effect of additional non-radiative energy transfer taking place upon SPI-MC isomerization. We obtain the lifetimes $\tau_{21} = 1.42$ ns and $\tau_{20} = 4.2$ ns from the fluorescence lifetime measurements (see Section S4 in the Supporting Information). As

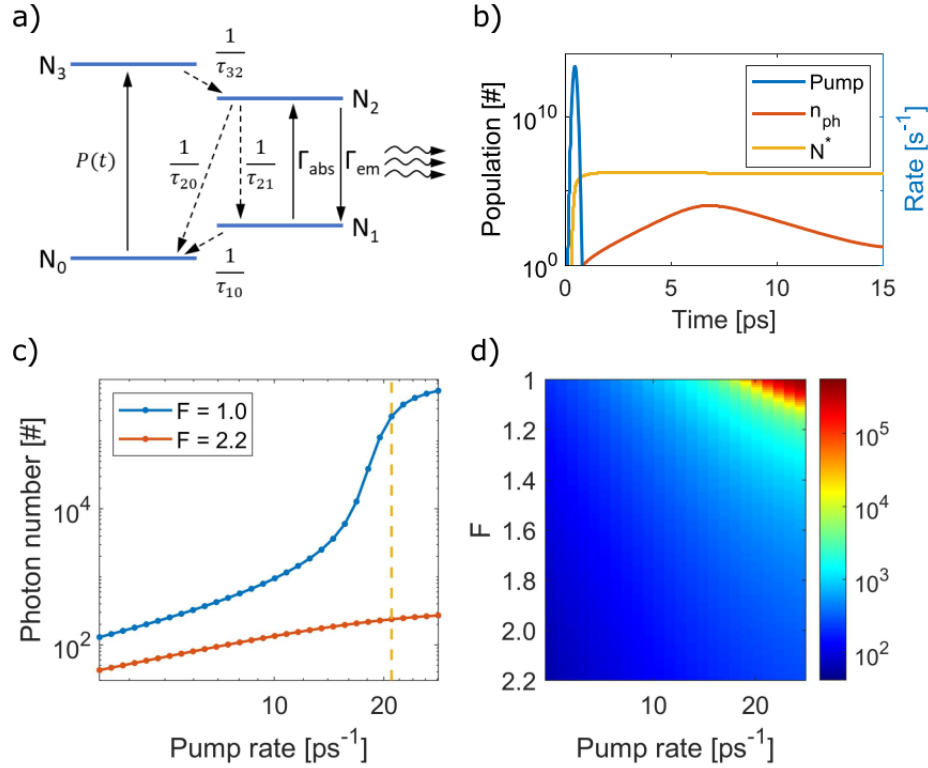


Figure 4: Rate equation simulation of photoswitching. (a) Schematic of the 4-level rate equation model. (b) Rate equation simulation showing the pump pulse (blue), population inversion $N^* = N_2 - N_1$ (yellow), and output lasing pulse n_{ph} (red solid line) as a function of time. The result is computed just above the lasing threshold at pump rate 21 ps^{-1} (yellow dashed line). The full width at half maximum of the output lasing pulse is 2.6 ps. (c) Photon number of the cavity mode, n_{ph} , as a function of pump rate for $F = 1.0$ and $F = 2.2$. (d) Photon number of the cavity mode, n_{ph} , as a function of pump rate and F .

vibrational relaxation times for coumarin dyes are typically on the order of 100 fs...1 ps,^{44,45} we set τ_{32} and τ_{10} to 0.5 ps. For the simulations we use 2×10^6 molecules and let the simulation run for a sufficiently long time so that all excitations created by the pump have decayed (20 ns).

To qualitatively model the effect of photoswitching, we modify the loss rate of the photon mode by a factor F , namely as $F \times n_{\text{ph}}/\tau_{\text{ph}}$ in Eq. (1). The factor F therefore accounts for the overall increase of photon loss from the lasing mode due to increased absorption imposed by switching the photochromic molecules from the SPI to the MC form. We map the factor F to experiments by reproducing the measured spontaneous emission intensities for both forms of the dye near the Γ -point. Switching the molecules to the absorbing form reduces the spontaneous emission intensity to approximately 30% of the initial value. In the simulation this is reproduced by increasing the loss factor F from 1.0 to 2.2. Fig. 4 (c) shows the pump dependence curves for $F = 1.0$ (SPI form of the molecule) and $F = 2.2$ (MC form of the molecule). In the spontaneous emission regime the ratio of photon numbers between $F = 2.2$ and $F = 1.0$ curves is around 30%. In the case of $F = 1.0$ the model reproduces the experimentally obtained lasing threshold curve. For $F = 2.2$, no onset of lasing is observed in the range of pump fluences studied. At pump rate 21 ps^{-1} (yellow dashed line), the ratio of output intensities between the two forms of the molecule is 10^3 .

Fig. 4 (d) presents the photon number as a function of pump rate and F . The simulation suggests that already a 20–30% increase in photon loss rate would enable 1–2 orders of magnitude modulation of the lasing signal. Simulations in the time domain, as exemplified by Fig. 4 (b), show that the lasing pulse comes out on a few picosecond timescale. This is in accordance with the previous studies in plasmonic lattices demonstrating ultrafast lasing pulse generation.⁴¹

Finally, in Figs. 5 (a-b) we measure the angle and energy resolved emission spectra at a fixed pump fluence of 0.19 mJcm^{-2} ($\sim 1.1 P_{\text{th}}$) after exposing the sample to (a) visible and (b) UV light (30 s exposure time). In Fig. 5 (a), the system exhibits lasing similar to

that observed in Fig. 3. After the UV exposure, however, the sample transits to spontaneous emission regime due to increased overall losses in the mode volume of the plasmonic nanoparticle lattice. Remarkably, by visible light exposure, the lasing action can be recovered. Figs. 5 (c-e) show the emission intensity at Γ -point ($k_y = 0$) as well as the linewidths and beam divergences for alternating UV and visible light exposures, demonstrating the repeatability of all-optical control of lasing in nanoparticle lattices. The switching is established over 2-3 orders of magnitude, which is in agreement with the rate equation simulation. We also observe approximately 100 and 50 fold modulation in linewidth and beam divergence, respectively.

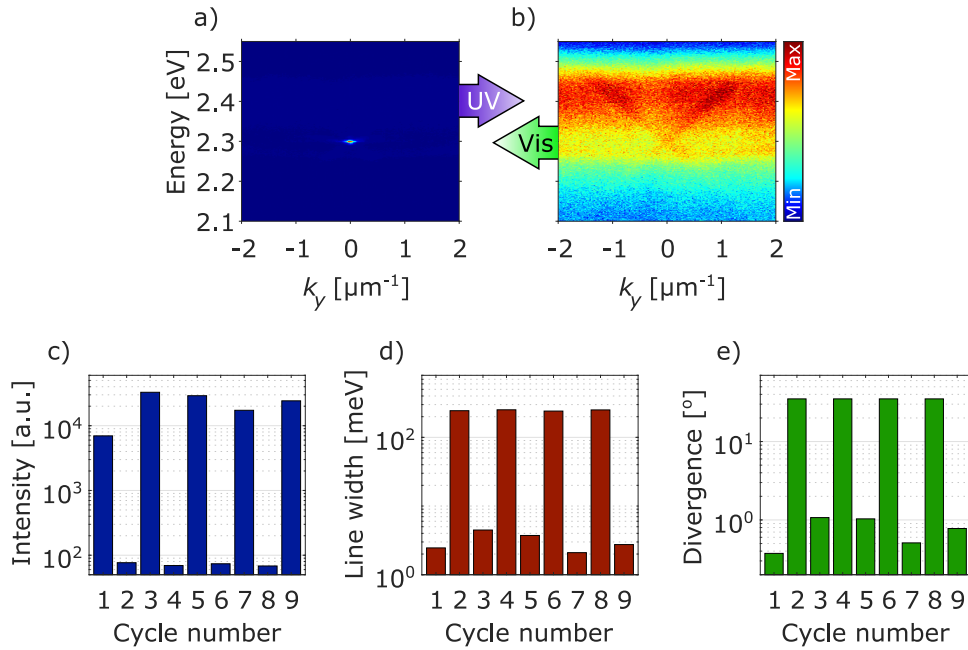


Figure 5: Optical control of lasing. (a-b) Normalized emission intensity of a photoswitchable lasing sample after (a) visible and (b) UV light illumination. (c) Integrated intensity values near the Γ -point, (d) measured emission linewidths (full width at half maximum) at $k_y = 0$ and (e) beam divergences after alternating UV and visible light exposures.

As concluded from comparing the sample emission for both forms of the photochromic dye, the overall increase in the photon loss rate causes the spontaneous emission intensity to drop to about 30% of the initial value when the photochromic molecules are switched to the MC form. Importantly, due to the highly nonlinear response of plasmonic lasers,

the additional losses caused by the SPI-MC switching are sufficient to produce orders of magnitude variation in the output intensity.

The results of the time domain simulations suggest that with the realistic parameters used in the model, the lasing dynamics occur on a picosecond timescale, in agreement with previous findings.^{41,46} Combining our system with photochromic compounds capable of switching their conformation on a 1–10 ps timescale (for instance, diarythelene molecules⁴⁷) could provide ultrafast, all-optical switching with exciting future prospects toward optoelectronic switches and memories that could operate at up to THz-frequencies.

We point out that the functional components of the presented system span over particularly wide range of different length scales (6 orders of magnitude). First, the composite gain medium is a mixture of 2 different molecules of approximately 1 nm size. Second, their (UV and visible light tunable) intermolecule (donor to acceptor) distances range from infinite to approximately 3 nm. Third, the plasmonic nanoparticle resonances have high near field enhancements within the range of 10 nm from the surface of the particle. Fourth, plasmonic nanoparticle lattices exhibit resonances having coherence lengths on the order of 10 μm due to radiative coupling between individual nanoparticles, and, finally, the lasing in such lattices exhibits coherence lengths in excess of 1 mm, which appears to be only limited by the lattice size or the pump area.⁴⁶ Very recently, the challenges of analysing light-matter interactions and energy transfer in such multiscale systems have been recognised and addressed by an elaborate study which combines classical (macroscopic) and quantum mechanical (microscopic) descriptions.⁴⁸ To understand the potential role of plasmon-enhanced energy transfer in our results, such mesoscopic approach should be combined with description of the non-linear dynamics caused by gain and losses. We view this challenge as a fertile ground for novel findings and a well justified motivation for further studies in such tunable multiscale systems.

Conclusions

To summarize, we have shown how photochromic molecules enable all-optical emission control and switching of lasing in nanoparticle arrays. By combining the organic dye with photochromic molecules we added a reversible, optically controllable loss mechanism to the system. Combined with the non-linear response of the plasmonic laser, it enables reversible switching between spontaneous emission and lasing regimes, with 2–3 orders of magnitude modulation in light output intensity. Our observations are qualitatively explained by a 4-level rate equation model. In addition to the strong intensity modulation, we observe 100 and 50 fold modulation in linewidth and beam divergence, respectively, when switching between the two regimes. While optical gain and lasing have been extensively investigated in plasmonic systems, the control over the losses has been scarcely studied. All-optical, reversible control provided by photochromic molecules over this quantity is particularly appealing as it opens the route towards actively tunable plasmonics, in time-scales limited only by the switching time of the molecules which can reach the THz scale.

Methods

Sample fabrication

Square arrays of silver nanoparticles were fabricated on borosilicate glass slides using standard electron beam lithography methods. The substrates were spin-coated with a polymethyl methacrylate (PMMA A4) resist layer which was solidified on a hot plate. The PMMA layer was patterned by exposing it to an electron beam and developed in 1:3 methyl isobutyl ketone : isopropanol solution. The nanoparticle diameter was 60 nm, the array size $100 \times 100 \mu\text{m}^2$ and the array periodicity 350 nm. A 2 nm titanium adhesion layer and a 30 nm silver layer were evaporated on the patterned PMMA layer, followed by lift-off. For transmission (lasing) measurements the nanoparticle arrays were immersed in index-matching oil (fluorescent gain

medium) and covered with a microscope cover slip.

Angle-resolved intensity spectroscopy

Angle-resolved intensity measurements were obtained by collecting light emitted by or transmitted through the sample with an objective (10x, 0.3 NA) whose back focal plane was focused to the entrance slit of a spectrometer. The angle of light exiting the sample θ_y is mapped on the spectrometer slit as $k_y = k_0 \sin \theta_y = 2\pi/\lambda_0 \sin \theta_y$ where λ_0 is the free space wavelength. Each point on the slit and likewise each pixel row on the charge-coupled device camera corresponds to a k_y value while the pixel columns are used to resolve the energy spectrum at each k_y such that $E = hc/\lambda_0$ where h is the Planck constant and c is the speed of light. An iris was used in the real image plane of the setup to filter out any light emerging from outside the array. In the lasing measurements the sample was pumped with femtosecond laser pulses (1 kHz repetition rate, 100 fs pulse duration, 450 nm center wavelength, 45° incident angle) with a flat pulse profile.

Time-resolved fluorescence spectroscopy

Fluorescence intensity decay curves were measured by using a time-correlated single photon counting (TCSPC) system (PicoQuant, GmBH) equipped with PicoHarp 300 controller and a PDL 800-B driver. LDH-P-C-483 laser head was used to excite the sample at 483 nm with a time resolution of ~ 110 ps. The fluorescence signal was detected by means of microchannel plate photomultiplier tube (Hamamatsu R2809U). Cutoff filter of transmission > 520 nm was applied to reduce the impact of the excitation light scattering. Fluorescence decays were monitored at 540 nm for all samples to maximize the signal. Instrumental response function was measured separately at monitoring wavelength 483 nm and used for deconvolution

analysis of the fluorescence decays followed by their fitting by sum of exponents

$$I(t, \lambda) = \sum_i a_i e^{-t/\tau_i}, \tag{6}$$

where τ_i is the fluorescence lifetime and a_i is the pre-exponential factor.

Spectrophotometric measurements

Spectrophotometric studies were performed with an Agilent Cary 60 spectrophotometer and a large custom-built cavity with an Ocean Optics qpod 2e temperature-controlled cuvette holder and a Prior Scientific Lumen-1600 light source. The light source includes LEDs of different wavelengths and the wavelengths used for switching the photochromic molecules were 365 nm (SPI to MC) and 550 nm (MC to SPI).

Simulations

The FDTD simulations were conducted using Lumerical’s FDTD Solutions package. The nanoparticle was modeled as a 60 nm diameter, 30 nm tall silver cylinder in the middle of the simulation volume using tabulated values for the permittivity.⁴⁹ The particle was surrounded by a 1 nm shell of a dielectric material with a refractive index of $1.54 + 0.2i$, while the background refractive index of the simulation was set to 1.54. The lossy shell was used to reduce meshing related artefacts at the silver boundary in the field profile. The particle was meshed using a 0.5 nm regular mesh, and the boundary conditions were anti-symmetric for x , symmetric for y , and perfectly matched layers for z , leaving only one quarter of the nanoparticle in the final simulation volume. The period in x and y was 350 nm, and the simulation height in z 2 μm . An x -polarized plane wave traveling along the z -axis was used to excite the system, and the field profile was recorded in the xy -plane at $z = 0$.

We use coupled dipole approximation (CDA) to compute the extinction spectrum of a two-dimensional periodic array of nanoparticles. See Ref.⁴⁰ and references therein for a

detailed description of the method. In CDA, each nanoparticle in the array is treated as a single electric dipole characterized by its induced polarization

$$\vec{P}_n = \alpha(\vec{E}_n^{\text{inc}} + \vec{E}_n^{\text{sc}}), \quad (7)$$

where \vec{E}_n^{inc} is the incident field acting on the n th particle, and \vec{E}_n^{sc} is the field resulting from scattering by all the other nanoparticles. The polarizability α is calculated for an ellipsoid nanoparticle under the modified long-wavelength approximation.^{40,50} The scattered field can be written as

$$\vec{E}_n^{\text{sc}} = \sum_{m=1, m \neq n}^N \underline{\vec{G}}(\vec{r}_n, \vec{r}_m) \vec{P}_m, \quad (8)$$

where $\underline{\vec{G}}$ is the electric-field dyadic Green's function ($R = |\vec{R}| = |\vec{r}_n - \vec{r}_m|$)

$$\underline{\vec{G}}(\vec{r}, \vec{r}') = \frac{e^{ikR}}{4\pi\epsilon_0 R} \left[\left(\underline{\vec{I}} - \frac{\vec{R}\vec{R}}{R^2} \right) + \frac{ikR - 1}{k^2 R^2} \left(\underline{\vec{I}} - 3\frac{\vec{R}\vec{R}}{R^2} \right) \right]. \quad (9)$$

The Green's function in Eq. (9) is singular at $\vec{r}_n = \vec{r}_m$. In the dipole approximation, it is conventional to regularize the Green's function by excluding the self-terms representing the interaction of the nanoparticle with itself.⁵¹ Substituting the scattered field in Eq. (8) into Eq. (7) leads to a linear system

$$\sum_{m=1}^N \underline{\vec{A}}_{nm} \vec{P}_m = \vec{E}_n^{\text{inc}}, \quad (10)$$

where the elements of the interaction matrix $\underline{\vec{A}}$ are given by

$$\underline{\vec{A}}_{nm} = \begin{cases} \underline{\vec{\alpha}}^{-1} \underline{\vec{I}}, & n = m, \\ -\underline{\vec{G}}(\vec{r}_n, \vec{r}_m), & n \neq m. \end{cases} \quad (11)$$

Note that whilst the self-terms of the Green's function are neglected, the interaction matrix $\underline{\vec{A}}$ contains the single-nanoparticle polarizabilities in the diagonal, representing the self-interaction of the nanoparticles. The induced polarization of the nanoparticles, obtained by

solving the linear system Eq. (10), are used to calculate the extinction cross section:⁵²

$$C_{\text{ext}} = \frac{4\pi k}{|E^{\text{inc}}|^2} \sum_{n=1}^N \text{Im} \left(\vec{E}_n^{\text{inc}*} \cdot \vec{P}_n \right). \quad (12)$$

Extinction efficiency is obtained from the cross section as $Q_{\text{ext}} = C_{\text{ext}}/A$, where A is the surface area of the nanoparticle array. The results shown in Fig. 3 (d) were obtained for 30×30 array of nanoparticles excited with a plane wave polarized in the x -direction. The simulation parameters were as in the experiments: periodicity was 350 nm and the particle radius and height were 30 nm. Refractive index of the background was 1.52. We used tabulated values for the permittivity of silver.⁴⁹

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Supporting Information

The Supporting Information is available free of charge at <http://pubs.acs.org>.

Photostationary state composition; Thermal lifetime of the MC form; Fluorescence lifetimes; Lifetimes in the rate-equation model; Schematic of the angle-resolved measurement setup; Sample emission linewidth as a function of pump fluence

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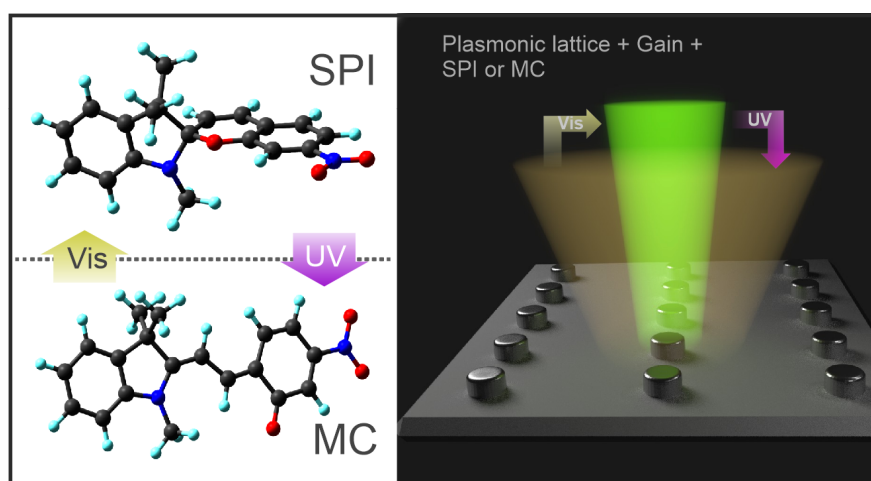
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For Table of Contents Use Only

Manuscript title: All-Optical Emission Control and Lasing in Plasmonic Lattices

Authors: Jani M. Taskinen, Antti J. Moilanen, Heikki Rekola, Kim Kuntze, Arri Priimagi, Päivi Törmä, and Tommi K. Hakala

Brief synopsis of the TOC-graphic: All-optical control of highly directional lasing is achieved by combining plasmonic nanoparticle lattices with an organic gain dye and photochromic molecules, which can be optically switched between two conformational states.



Supporting Information – All-Optical Emission Control and Lasing in Plasmonic Lattices

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Supporting Tables: S1-S2

1. Photostationary state composition

The absorption spectra of a 53 μM indolinospiropyran solution in a 1 cm cuvette upon illumination with 365 and 550 nm are shown in Fig. S1. Upon illumination with 550 nm, the molecules are purely in the SPI form, whereas upon illumination with 365 nm, a photostationary state (PSS) with both SPI and MC forms is reached (see Fig. S1 (a)). Due to the spectral shape of the isomers, the PSS composition cannot be deduced from this data only.

However, the composition can be approximated by comparing the measured molar extinction coefficient of the MC absorption band with a literature value for the pure isomer. In polar protic solvents, ϵ_{MC} can be approximated to be $5.2 \times 10^4 \text{ M}^{-1}\text{cm}^{-1}$.¹ From this value, the PSS composition can be estimated to contain 14% MC and 86% SPI. However, no specific value was available for benzyl alcohol and the approximation should thus be treated with caution. Similar spectra were acquired between a microscope slide and a cover glass (7.0 μl of 40 mM solution between the glass slides). The molar absorptivity cannot be measured as the precise optical length is not known, but based on the relative intensity of the peaks at *ca.* 350 and 550 nm, the PSS composition is very close to that observed in the more diluted solution.

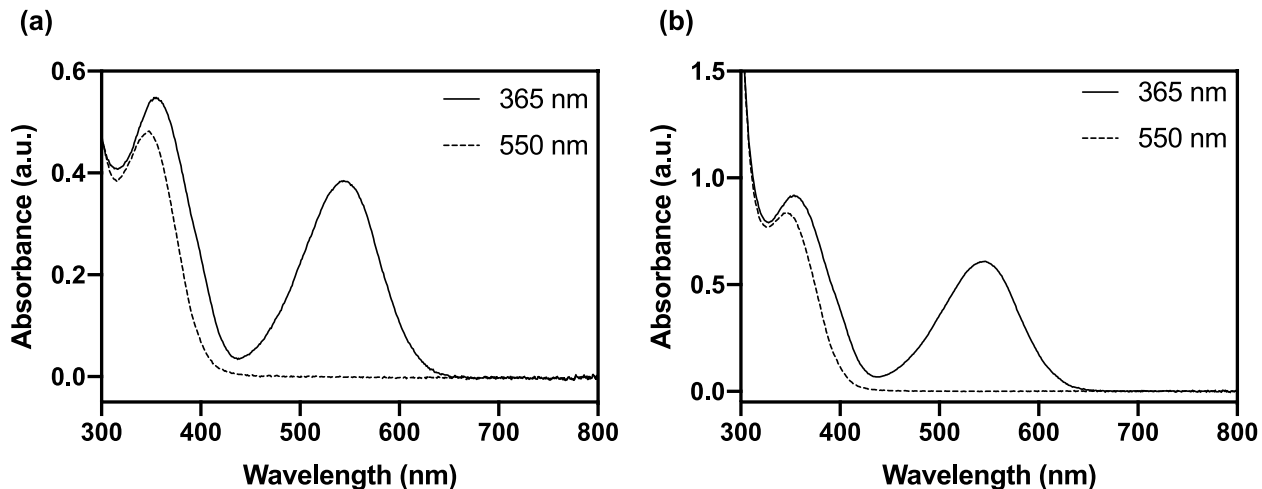


Figure S1: The absorption spectrum of the pure SPI isomer (dashed line) and the photostationary mixture of SPI and MC (solid line) in (a) 53 μM benzyl alcohol solution in a 1 cm quartz cuvette and (b) 40 mM benzyl alcohol solution between a microscope slide and a cover glass.

2. Thermal lifetime of the MC form

Thermal relaxation from the MC form back to the pure SPI was studied in both solutions at room temperature, yielding a good monoexponential fit in both cases. Interestingly, the lifetime of the relaxation was significantly longer for the more concentrated sample. It should

be noted that the relaxation seems to end up in a mixture of SPI and MC with *ca.* 3% MC with the earlier approximation, meaning that the isomerization is not fully reversed with heat. The monoexponential decay fits shown in Fig. S2 are of the form

$$A(t) = A_{\text{inf}} + (A_0 - A_{\text{inf}})e^{-t/\tau}, \quad (\text{S1})$$

where t is time, A_0 is the absorbance at $t = 0$, A_{inf} is the lowest absorbance reached *via* thermal relaxation and τ is the lifetime of the relaxation. Table S1 shows the thermal relaxation parameters for both solutions.

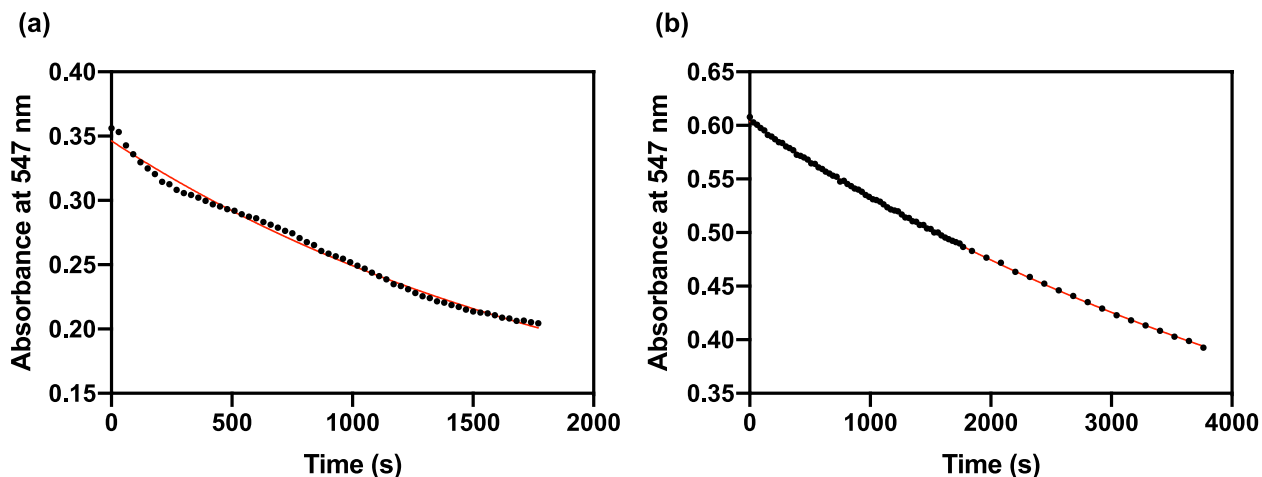


Figure S2: Measured absorbance of indolinospiropyran at 547 nm upon thermal relaxation from PSS to thermal equilibrium (black dots) and an exponential decay fit (red line) for (a) 53 μM and (b) 40 mM in benzyl alcohol solution.

Table S1: Parameters for the thermal relaxation (Eq. S1) from PSS after illumination with 365 nm. Here $\tau_{1/2}$ is the half-life of the fluorescence.

	$\tau_{1/2}(\text{s})$	$\tau(\text{s})$	A_0	A_{inf}	R^2
53 μM	1424	2055	0.3464	0.0945	0.9928
40 mM	3767	5435	0.6037	0.1840	0.9997

3. Fluorescence lifetimes

Fluorescence decay curves were measured for pure C6 as well as a mixture of C6 (30 mM) and indolinospiropyran (40 mM) after irradiation with 550 nm and 365 nm (SPI and MC forms). The fluorescence lifetime of pure C6 was measured in a more diluted solution (*ca.* 100 μ M). The fluorescence intensity decay curves are presented in Fig. S3. The decays of diluted C6, C6 + SPI and C6 + MC were fitted with bi- or triexponential functions (Table S2, Fig. S4).

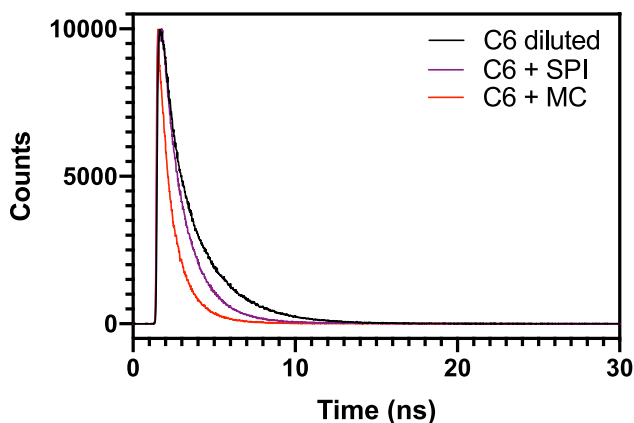


Figure S3: Fluorescence intensity decay of C6 at *ca.* 100 μ M concentration, and C6 (30 mM) + indolinospiropyran (40 mM) upon illumination with 550 nm (SPI) and 365 nm (MC).

Table S2: Lifetimes obtained by multiexponential fitting to the measured fluorescence lifetimes shown in Figs. S4 (a-c). Also shown are the pre-exponential factors (amplitudes) of each fitting in brackets.

	τ_1	τ_2	τ_3
C6 <i>ca.</i> 100 μ M	2.36 ns (0.0816)	-	0.41 ns (0.0541)
C6 + SPI	1.42 ns (0.1072)	5.51 ns (0.0016)	0.26 ns (0.0231)
C6 + MC	1.06 ns (0.0796)	5.17 ns (0.0006)	0.14 ns (0.0977)

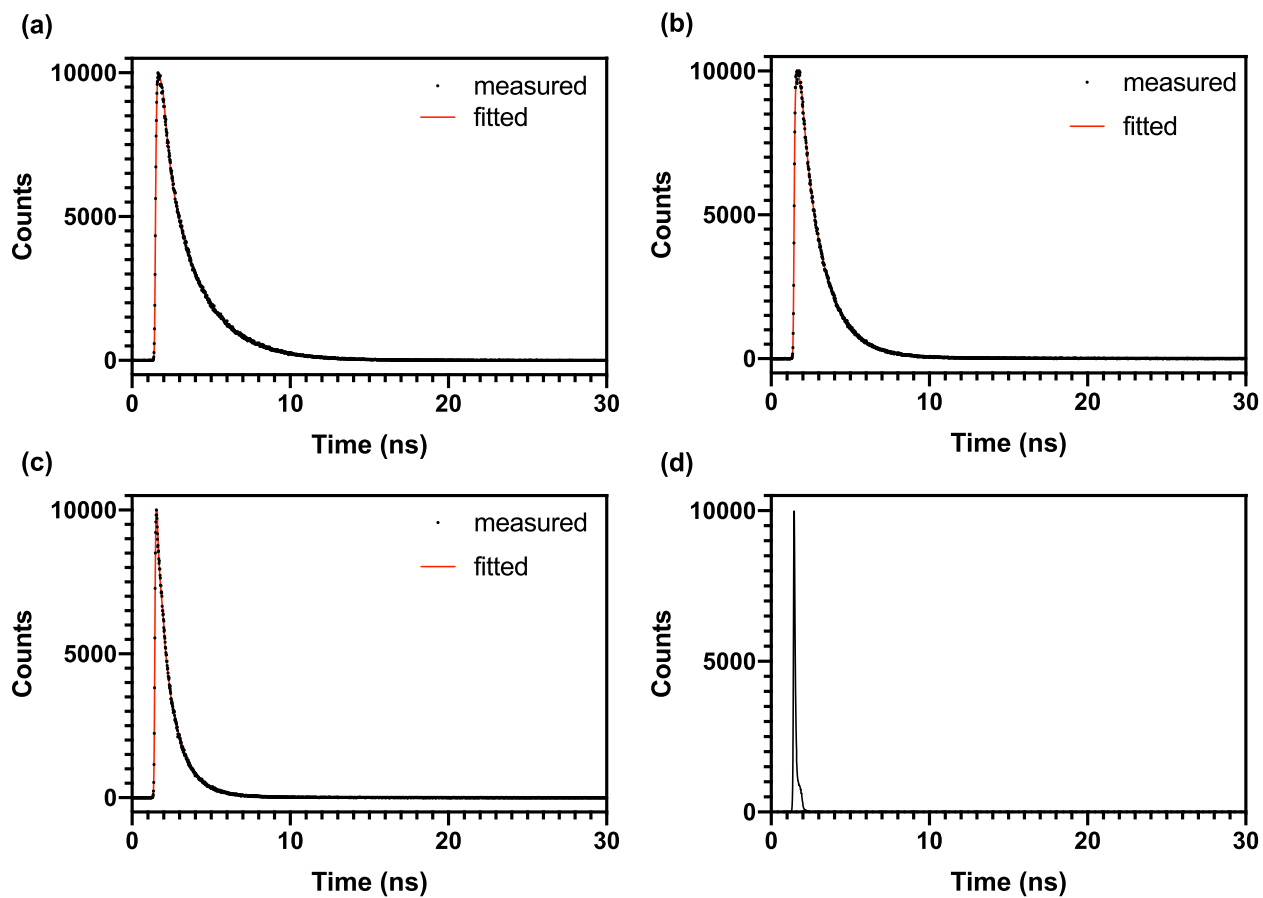


Figure S4: Measured and fitted fluorescence decay curves for (a) *ca.* 100 μ M C6, (b) C6 + SPI and (c) C6 + MC, and (d) instrument response function.

4. Lifetimes in the rate-equation model

We obtain the radiative and non-radiative lifetimes from the fluorescence lifetime measurement shown in manuscript Fig. 2 (c). The decrease of the effective fluorescence lifetime from 1.42 ns to 1.06 ns upon SPI-MC isomerization is attributed to the increased non-radiative energy transfer between the compounds. We can write the following relations between the radiative (rad) and non-radiative (nrad) lifetimes of the dye in the SPI and MC forms:

$$\frac{1}{\tau_{\text{C6+SPI,rad}}} + \frac{1}{\tau_{\text{C6+SPI,nrad}}} = \frac{1}{1.42} \quad (\text{S2})$$

$$\frac{1}{\tau_{\text{C6+MC,rad}}} + \frac{1}{\tau_{\text{C6+MC,nrad}}} = \frac{1}{1.06}. \quad (\text{S3})$$

Assuming the SPI-MC isomerization does not directly affect the radiative spontaneous emission lifetime, we can write $\tau_{\text{C6+SPI,rad}} = \tau_{\text{C6+MC,rad}}$ and denote this lifetime as τ_{21} . Further, taking that the intermolecular non-radiative energy transfer is negligible in the case of non-absorbing form, $1/\tau_{\text{C6+SPI,nrad}} \approx 0$, we arrive to

$$\frac{1}{\tau_{21}} = \frac{1}{1.42} \quad (\text{S4})$$

$$\frac{1}{\tau_{21}} + \frac{1}{\tau_{20}} = \frac{1}{1.06}. \quad (\text{S5})$$

From here we obtain $\tau_{21} = 1.42$ ns and $\tau_{20} = 4.2$ ns. When modelling the photoswitching, τ_{20} is changed linearly from 4.2 ns (MC form) to a large value 10 ns (SPI form). The upper value 10 ns is chosen sufficiently long such that it has negligible effect on the system dynamics. We note that, overall, the values of τ_{20} are so large compared to other timescales of the system that the parameter has very little effect on the simulation results.

5. Schematic of the angle-resolved measurement setup

A schematic of the experimental setup used in the angle- and wavelength-resolved intensity measurements is shown in Fig. S5. A pump iris and a pump lens are used to image a flat pump spot to the sample. An infinity-corrected objective and a corresponding tube-lens form a magnified image of the sample at the real space iris, which spatially restricts the setup to only image the sample array. Finally, a k-space lens is used to form a Fourier image of the sample at the spectrometer slit. A halogen lamp is used to both illuminate the sample for the real space camera and function as a light source in transmission measurements.

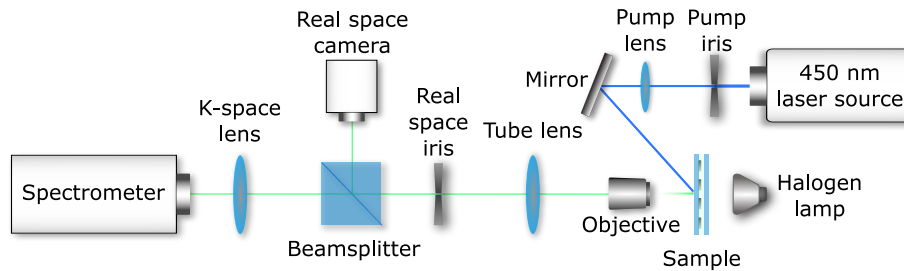


Figure S5: Schematic of the experimental setup used to obtain the angle- and wavelength-resolved intensity data.

6. Sample emission line width as a function of pump fluence

Fig. S6 shows the emission line width of the lasing sample as a function of pump fluence measured from the data set presented in Fig. 3 (a-b). The lower limit of the measured line widths was instrument-limited and shows slightly higher values compared to the data shown in Fig. 5 (d).

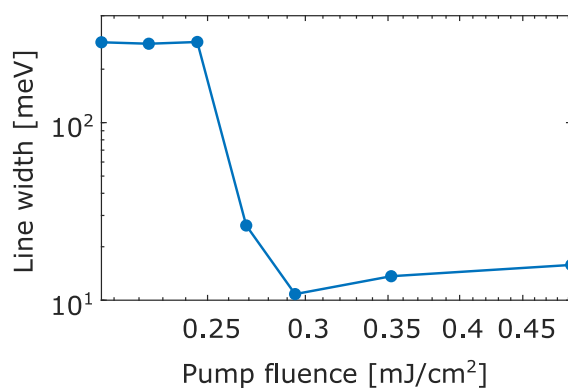


Figure S6: Emission line width (FWHM) of the lasing sample as a function of pump fluence.

Supplementary References

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