- 1 Microbiological, chemical and sensory spoilage analysis of raw Atlantic cod (Gadus
- 2 *morhua*) stored under modified atmospheres
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## 23 Abstract

24 During fish spoilage, microbial metabolism leads to the production of volatile organic 25 compounds (VOCs), characteristic off-odors and eventual consumer rejection. The aim of the 26 present study was to contribute to the development of intelligent packaging technologies by 27 identifying and quantifying VOCs that indicate spoilage of raw Atlantic cod (Gadus morhua) under atmospheres (%v/v CO<sub>2</sub>/O<sub>2</sub>/N<sub>2</sub>) 60/40/0, 60/5/35 and air. Spoilage was examined by 28 microbiological, chemical and sensory analyses over storage time at 4 or 8 °C. Selected-ion 29 flow-tube mass spectrometry (SIFT-MS) was used for quantifying selected VOCs and amplicon 30 sequencing of the 16S rRNA gene was used for the characterization of the cod microbiota. OTUs 31 32 classified within the *Photobacterium* genus increased in relative abundance over time under all 33 storage conditions, suggesting that *Photobacterium* contributed to spoilage and VOC production. The onset of exponential VOC concentration increase and sensory rejection occurred at high 34 35 total plate counts (7-7.5 log). Monitoring of early spoilage thus calls for sensitivity for low VOC 36 concentrations.

## 37 Keywords

38 Amplicon sequencing; *Photobacterium*; SIFT-MS; sensor; volatile organic compound

#### 39 **1. Introduction**

Raw fish is highly perishable due to the intrinsic properties of the product and inevitable 40 microbial activity. Spoilage of fish is primarily caused by microbial growth and metabolism and 41 is characterized by changes in the sensory properties that lead to unacceptable product quality 42 43 (Gram & Huss, 1996; Gram & Dalgaard, 2002; Gram et al., 2002). Shelf life of fish is affected 44 by several factors, including storage temperature, fish species, initial microbial contamination and packaging conditions (Sivertsvik, Jeksrud, & Rosnes, 2002). Even though  $10^7$  CFU/g has 45 46 generally been considered as a maximum acceptable microbial load for fish (Stannard, 1997), sensory rejection has typically been found at microbial levels between  $10^{6}$ - $10^{9}$  CFU/g (Dalgaard, 47 48 Mejlholm, Christiansen, & Huss, 1997; Mikš-Krajnik, Yoon, Ukuku, & Yuk, 2016; Nuin et al., 49 2008; Parlapani, Mallouchos, Haroutounian, & Boziaris, 2014; Parlapani, Verdos, Haroutounian, & Boziaris, 2015). 50

51 Specific spoilage organisms (SSOs) typically constitute a fraction of the initial microbiota and their outgrowth eventually leads to unacceptable changes in the product quality (Gram & 52 53 Dalgaard, 2002). The microbiota of fresh marine fish generally consists of psychrotrophic Gram-54 negative rod-shaped bacteria along with Gram-positive microbes (Gram & Huss, 1996). In 55 marine fish stored under refrigerated aerobic conditions, *Pseudomonas* and *Shewanella* spp. have 56 been observed to be dominating (Gram, Trolle, & Huss, 1987; Gram & Huss, 1996; Gram & Dalgaard, 2002; Vogel, Venkateswaran, Satomi, & Gram, 2005), whereas Photobacterium 57 58 phosphoreum has been identified as an SSO of Atlantic cod (Gadus morhua) under different modified atmosphere packaging (MAP) conditions (Dalgaard et al., 1997; Dalgaard, 1995; 59 60 Debevere & Boskou, 1996).

61 Odor is one of the most important quality determinants for fish freshness (Olafsdottir, Jonsdottir, Lauzon, Luten, & Kristbergsson, 2005). As a result of microbial metabolism, volatile organic 62 compounds (VOCs) are often produced, which leads to the production of characteristic off-odors 63 and off-flavors. Typical compounds associated with fish spoilage include acids, alcohols, 64 aldehydes, amines, ketones and sulfides (Gram & Dalgaard, 2002). The spoilage potential of 65 66 SSOs is characterized by their qualitative ability to produce off-odors, whereas spoilage activity refers to the quantitative ability to produce spoilage metabolites (Gram & Dalgaard, 2002; Gram 67 et al., 2002). Thus, evolution of spoilage-related VOCs could be used for fish quality evaluation 68 69 during storage. Different approaches for characterizing the VOC profile have been applied to marine fish species such as cod (Fernández-Segovia, Escriche, Gómez-Sintes, Fuentes, & Serra, 70 2006; Noseda et al., 2010), salmon (Jónsdóttir, Ólafsdóttir, Chanie, & Haugen, 2008; Jørgensen, 71 Huss, & Dalgaard, 2001; Macé et al., 2013; Mikš-Krajnik et al., 2016), sea bream (Parlapani et 72 al., 2014; Parlapani et al., 2015; Soncin, Chiesa, Panseri, Biondi, & Cantoni, 2009), sea bass 73 (Parlapani, Haroutounian, Nychas, & Boziaris, 2015), hake (Baixas-Nogueras, Bover-Cid, Vidal-74 Carou, Veciana-Nogués, & Mariné-Font, 2001), mackerel (Alfaro, Hernández, Baliño-Zuazo, & 75 Barranco, 2013) and turbot (Xu et al., 2014). 76

Intelligent packaging technologies aim at improving the quality and safety of the packaged
product and/or informing about its status by detecting, sensing, communicating, recording or
applying another intelligent function (Yam, Takhistov, & Miltz, 2005). Among these
technologies, sensors that convert physical or chemical information into an informative signal
have been considered to have high potential for future applications (Ghaani, Cozzolino, Castelli,
& Farris, 2016; Kerry, O'Grady, & Hogan, 2006; Vanderroost, Ragaert, Devlieghere, & De
Meulenaer, 2014). The use of sensor technologies for monitoring VOCs indicating fish spoilage

84	could enhance the detection of spoilage in individual packages, thus improving quality
85	evaluation and reducing food and packaging material waste throughout the supply chain. Even
86	though different applications for sensor-based quality monitoring of fish have been examined
87	(Bhadra, Narvaez, Thomson, & Bridges, 2015; Chung, Le, Tran, & Nguyen, 2017; Efremenko &
88	Mirsky, 2017; García et al., 2017; Morsy et al., 2016; Pacquit et al., 2006; Pacquit et al., 2007;
89	Perera, Pardo, Barrettino, Hierlermann, & Marco, 2010), there is still a limited number of studies
90	focusing on direct and real-time quantification of the VOC profile produced in the package
91	headspace during storage time, aiming at the development of intelligent packaging technologies.
92	Efficient quality monitoring of fish spoilage calls for fast, non-destructive and sensitive methods.
93	However, conventional quality analyses of fish packaged under modified atmospheres (MAs) are
94	commonly destructive and time consuming, such as the determination of total volatile basic
95	nitrogen (TVB-N) by steam distillation (Pacquit et al., 2006) or plate counts. Several
96	technologies have been used for rapid and accurate characterization of VOCs, including gas
97	chromatography-mass spectrometry (GC-MS) (Béné, Hayman, Reynard, Luisier, & Villettaz,
98	2001; G. Duflos et al., 2010; Edirisinghe, Graffham, & Taylor, 2007; Fernández-Segovia et al.,
99	2006; Grimm, Lloyd, Batista, & Zimba, 2000; Jaffrès et al., 2011; Leduc et al., 2012; Mikš-
100	Krajnik et al., 2016; Z. Zhang, Li, Luo, & Chen, 2010) and electronic noses (Natale et al., 2001;
101	Olafsdottir et al., 2005; Zaragozá et al., 2014). On the other hand, selective-ion flow-tube mass
102	spectrometry (SIFT-MS) can be used for non-destructive and sensitive real-time quantification of
103	VOCs from the package headspace. The technology is based on reactions between precursor ions
104	$(H_3O^+, NO^+, O^+)$ and target compounds, followed by the quantification of the resulting product
105	ions on the basis of their mass to charge (m/z) ratio. SIFT-MS has previously been validated for
106	fish metabolite research (Noseda et al., 2010) and used for VOC analysis of different food

107	products, including seafood (Noseda et al., 2012), meat (Carrapiso et al., 2015; Olivares,
108	Dryahina, Španěl, & Flores, 2012), fruit (Zhang et al., 2013; Zhang, Samapundo, Pothakos,
109	Sürengil, & Devlieghere, 2013; Zhang et al., 2014) and cheese (Castada, Wick, Taylor, &
110	Harper, 2014; Castada, Wick, Harper, & Barringer, 2015; Langford et al., 2012).
111	Identification and quantification of VOCs related to spoilage is of high importance for the
112	development of food quality monitoring. Establishing a relation between VOC production,
113	microbial growth (both in total amount and in specific microorganisms) and sensorial quality is
114	needed as the basis for the development of intelligent packaging solutions. In the present study,
115	spoilage of Atlantic cod packaged under modified atmospheres was analyzed by following
116	microbial growth, VOC concentrations and sensory quality during refrigerated storage. SIFT-MS
117	was used for the real-time quantification of VOCs from the package headspace and amplicon
118	(NGS) sequencing of the 16S rRNA gene was used for the characterization of the cod microbiota
119	at different stages of storage. The results of the present study contribute to the development of
120	intelligent packaging technologies within the CheckPack project (VLAIO grant number 130036).

## 121 **2.** Materials and methods

#### 122 2.1. Raw material

For each individual storage experiment, Atlantic cod (minimum body weight ca. 4.5 kg) was
caught in the North Atlantic Ocean (FAO zone 27), gutted, filleted and stored under ice. The fish
was transported to Belgium by air and delivered to the Laboratory of Food Microbiology and
Food Preservation (LFMFP) in polystyrene boxes under ice.

# 127 2.2. Packaging and storage

128 Cod fillet portions  $(217 \pm 5 \text{ g})$  were packaged under different atmospheres with a gas-product 129 ratio 2:1 using a tray sealer MECA 900 (DecaTechnic, Herentals, Belgium), multilayer packaging trays (PP/EVOH/PP, oxygen transmission rate 0.03 cm<sup>3</sup>/tray\*24h at 23 °C and 50 % 130 R.H.) and top film (PA/EVOH/PA/PP, oxygen transmission rate 6.57 cm<sup>3</sup>/m<sup>2</sup>\*24h\*atm at 23 °C, 131 50 % R.H. and 1 atm). Three different atmospheres and two storage temperatures were applied 132 (Table 1): independent batches of fish were used for each of the five storage experiments. In the 133 134 present study, the storage experiments are referred to as H4, H8, L4, L8 and Air, where the 135 notation of the MA conditions indicates high (H) or low (L) oxygen content and temperature in Celsius degrees (4 or 8). For the determination of background concentrations possibly 136 137 originating from the packaging materials and/or heat sealing, sample-free packages (blanks) with 138 similar gas atmospheres were prepared. The packages were stored at  $(4.0 \pm 0.7)$  or  $(8.0 \pm 0.4)$  °C until the day of analysis. On a regular basis, three randomly selected packages were analyzed. 139 140 After sampling, the remaining fish portion was packaged under vacuum using high barrier film bags (oxygen transmission rate  $< 2.7 \text{ cm}^3/\text{m}^2*24\text{h*bar}$  at 23 °C and 0 % R.H.) and stored at -32 141 °C for no longer than 120 days (sensory evaluation) or one year (amplicon sequencing). 142

# 143 2.3. Microbiological analysis

For microbiological analysis, 30 ± 0.1 g of individual fillet was aseptically weighed into a sterile
stomacher bag and diluted ten times in physiological saline peptone solution (PPS; 0.85 % m/v
NaCl, 0.1 % m/v peptone). The samples were homogenized in Stomacher Lab Blender (LED
Techno, Heusden-Zolder, Belgium) for one minute and appropriate decimal dilutions were
prepared in PPS. The total psychrotrophic count (TPC) was determined on Marine Agar (MA;
Difco Le Pont de Claix, France) by spread plating, lactic acid bacteria (LAB) on Man Rogosa
Sharpe Agar (MRS; Oxoid, Hamsphire, UK) or modified MRS (mMRS; yeast extract 4.0 g/L,

151	Lab-Lemco powder 8.0 g/L, peptone 10.0 g/L, sorbitan mono-oleate (Tween 80) 1 ml/L,
152	dipotassium hydrogen phosphate 2 g/L, sodium acetate 5 g/L, triammonium citrate 2 g/L,
153	magnesium sulphate 0.2 g/L, manganese sulphate 0.05 g/L; pH 8.6 at 25 °C; 20 % glucose
154	solution 100 mL/L after autoclaving) by pour plating, hydrogen sulfide (H <sub>2</sub> S) producers on Iron
155	Agar Lyngby (IAL; Oxoid) supplemented with L-cysteine (Fluka, Steinheim, Germany) by pour
156	plating, Pseudomonads on Pseudomonas Agar (PA; Oxoid) supplemented with Pseudomonas
157	CFC supplement SR 103E (Oxoid) by spread plating and Brochothrix thermosphacta on
158	Streptomycin Sulfate Thallous Acetate Actidione Agar (STAA; Oxoid) supplemented with
159	selective supplement SR 151E (Oxoid) by spread plating. Plates were incubated at 22 °C for 2

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160 (PA and STAA), 3 (MRS and IAL) or 5 days (MA).

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# 161 2.4. Quantification of spoilage related VOCs by SIFT-MS

A selected-ion flow-tube mass spectrometer (Voice 200, Syft Technologies<sup>TM</sup>, Christchurch, 162 New Zealand) was used for quantifying a predefined set of VOCs in the package headspace. 163 Principles of the instrument have been described elsewhere (Noseda et al., 2010). The 164 165 compounds (Table 2) were selected on the basis of previous research and literature survey. The package headspace was sampled with a flow rate of 32 ml/min during 60 seconds (preparation 166 10s, sample 50s) through a septum inserted on the package lid and the VOC concentrations were 167 168 averaged over eleven data points. Two consecutive gas samples per package were analyzed. During sampling, the headspace was connected to atmospheric air with a needle inlet in order to 169 170 avoid package collapse and subsequent change in the internal conditions of the package. Respectively, empty packages (blanks, n=9-14) from each headspace-temperature combination 171 were randomly analyzed throughout the storage time and used for determining the limit of 172

173 quantification (LOQ) of each compound and for background subtraction. Concentrations of the

174 VOCs were determined with the LabSyft software (Syft Technologies<sup>TM</sup>).

The relative standard deviation (SD%) of each VOC concentration during a SIFT-MS measurement
was calculated as

177 
$$SD_{\%} = SD_m/x_m * 100 \%$$
 (1)

where  $x_m$  is the average and  $SD_m$  the standard deviation of a single SIFT-MS measurement (n=11). VOCs with concentrations exceeding 25 % average relative standard deviation during the entire storage time within a certain packaging condition were excluded from further analysis.

The Limit of Quantification (LOQ) was calculated with the International Union of Pure and
Applied Chemistry (IUPAC) equation (Mocak, Bond, Mitchell, & Scollary, 1997):

183 
$$LOQ = x_{bl} + 6*SD_{bl}$$
 (2)

where  $x_{bl}$  is the total average and  $SD_{bl}$  the standard deviation of the blanks. Background was subtracted from the measured concentrations that exceeded the LOQ: the reported results are measured concentrations minus  $x_{bl}$ .

# 187 2.5. Headspace composition (% CO<sub>2</sub>/O<sub>2</sub>), pH and color measurements

The headspace gas composition (% v/v CO<sub>2</sub>/O<sub>2</sub>) was analyzed with a gas analyzer (CheckMate<sup>®</sup> 9900 CO<sub>2</sub>/O<sub>2</sub>, Dansensor A/S, Ringsted, Denmark). pH was determined as an average of three consecutive measurements from randomly selected spots in individual fillets within 30 minutes after opening the package using a pH electrode (Lab<sup>®</sup> 427, Mettler Toledo GmbH, Schwerzenbach, Switzerland) connected to a pH meter (SevenEasy, Mettler Toledo GmbH). The product color was determined as an average of ten measurements from randomly selected fillet spots by a spectrophotometer (CM 2500d, Konica Minolta Sensing Inc., New Jersey, USA) and related SpectraMagic<sup>TM</sup> NX color data software. Color was measured through a small Petri dish (diameter 230 mm) using the CIE  $L^* a^* b^*$  color space with a standard 10° observer and Illuminant D65.

#### 197 2.6. Sensory evaluation

Sensory evaluation was based on olfactory evaluation and performed in individual booths under red light (UGent Sensolab). A panel of 8-12 persons having experience in sensory evaluation of fish was formed from the laboratory staff at LFMFP. One out of three daily replicates (A-C) was randomly selected and used per testing session. The samples were thawed at 2 °C overnight, cut to  $5.0 \pm 0.1$  g portions and presented to the panelists at 4 °C in odor-free, transparent plastic cups (diameter 67 mm; AVA, Temse, Belgium), closed with lids (AVA) and labelled with three-digit random codes generated with Excel 2013 (Windows).

Ranking tests (ISO, 2006) were used to determine if significant differences occurred between different stages of storage within a certain packaging condition. Four samples were presented to the panelists to be ranked from least fresh (1) to most fresh (4). For conditions H4, H8, and L4, a second ranking test was performed for the critical days identified by the first test. The collected data was subjected to a Friedman test followed by a Least Significant Difference test (Excel 2013 for Windows) in order to determine whether significant differences occurred between samples from different days of storage.

Acceptance tests were used to determine the quality of cod samples from different stages of storage within a certain packaging condition. Four samples were presented to the panelists along with a fresh reference (day 0) from the same lot. A five-point scale (very good, good, satisfactory, marginal, spoiled) was used for the evaluation.

# 216 2.7. Amplicon sequencing

217 16s rRNA gene amplicon sequencing analysis was used for the characterization of the cod 218 microbiota over storage time. Three samples stored at -32 °C were selected to represent early, 219 intermediate and late stages of storage (Table 1). One randomly selected sample out of three daily 220 replicates (A-C) was used for the analysis.

A phenol/chloroform extraction procedure with mechanical disruption using a FastPrep device 221 222 (Vilchez-Vargas et al., 2013) was used for the extraction of DNA. Bacterial cells were aseptically collected from the frozen sample surface by swabbing. An individual swab was placed in an 223 Eppendorf tube with 200 mg glass beads and 1000 µl of lysis buffer (100 mM Tris; 100 mM 224 EDTA; 100 mM NaCl; 1 wt/vol % polyvinylpyrrolidone; 2 wt/vol % sodium dodecyl sulphate; 50 225 ml water; pH 8). The tube was transferred to the FastPrep-24 instrument (MP Biomedicals, Santa 226 227 Ana, California, USA) and disrupted twice at 1400 rpm for 60 s. After centrifuging at maximum speed for 5 min, phenol-chloroform-isoamilic alcohol (500 µl; pH 7) was added to the supernatant 228 and the solution was thoroughly vortexed and centrifuged at maximum speed for 60 s. Chloroform 229 230 (700 µl) was added to the supernatant, mixed by vortexing and centrifuged at maximum speed for one minute. The resulting upper phase was divided into two Eppendorf tubes (450 µl per tube) 231 232 where sodium acetate (3M; 45 µl) was added, followed by mixing and addition of isopropyl alcohol (-20 °C; 500 µl). The solution was mixed by inverting, stored for one hour at -20 °C and 233 centrifuged at maximum speed for one minute at 4 °C. The resulting pellet was dried and dissolved 234 235 into T10E1 (100 µl).

Library preparation and sequencing was carried out at LGC Genomics (Germany) according to

the procedure presented by De Vrieze et al. (2016). The PCR mix contained 1 ng of DNA extract

and PCRs showing low yields were further amplified for 5 additional cycles if needed.

Sequencing was done on an Illumina MiSeq platform using v3 Chemistry (Illumina, San Diego,
California, USA) along with a mock community that was included in triplicate in the sequencing
run to assess the sequencing quality. The mock community consisted of the genomic DNA of 12
species from 10 different phyla and was pooled to an equimolar concentration of 16S rRNA gene
copies based on Q-PCR with the Illumina primers.

The mothur software package v. 1.38.0 (Schloss et al., 2009) and guidelines developed by P.

245 Schloss (Miseq sop.12th October 2016; Kozich, Westcott, Baxter, Highlander, & Schloss, 2013)

246 were used for processing the amplicon sequencing data. From the total number of forward and

reverse reads, contigs with lengths outside of the 2.5 - 97.5 % quantiles or sequences with

ambiguous base calls were removed. Remaining unique sequences were aligned to the mother-

reconstructed SILVA Seed alignment v. 123 (Pruesse et al., 2007). Unique sequences were pre-

clustered within a distance of 1/100 nucleotides and chimeras were screened with UCHIME

251 (Edgar, Haas, Clemente, Quince, & Knight, 2011). Next, sequences were classified using RDP v.

Non-bacterial or unidentified sequences were removed and the remaining OTUs were clustered
using average linkage and 97 % sequence identity. Single-read OTUs were considered as likely
errors and discarded from further analyses. The alpha diversity was examined by rarefaction
curves and community richness estimators Chao1984 (Chao, 1984), ChaoBunge2002 (Chao &
Bunge, 2002) and ACE-1 (Chao & Lee, 1992), diversity estimators Shannon (Shannon &
Weaver, 1949), Simpson (Simpson, 1949) and inverse Simpson, and evenness estimator Pielou

259 (Pielou, 1966).

#### 260 **3. Results**

<sup>252 14 (</sup>Cole et al., 2009) and Wang's algorithm.

## 261 3.1. Headspace composition (% $CO_2/O_2$ )

The development of headspace gas concentrations  $(CO_2/O_2)$  is presented in Table 3. Under high O<sub>2</sub> concentrations (H4 and H8), simultaneously, an initial increase in oxygen levels and a decrease in carbon dioxide levels were observed, the latter likely due to the dissolution of carbon dioxide into the food product. Under all tested conditions, oxygen content decreased and carbon dioxide content subsequently increased after several days of storage at the time of progressing microbial growth (see 3.3.).

## 268 3.2. *pH and color*

The evolution of pH and color variables L\*, a\* and b\* is presented in Table 3. Throughout storage time, pH was  $6.33 \pm 0.12$ ,  $6.57 \pm 0.16$ ,  $6.53 \pm 0.17$ ,  $6.68 \pm 0.15$  and  $6.70 \pm 0.05$  under the conditions H4, H8, L4, L8 and Air, respectively. In addition to some increase in yellowness (b<sup>\*</sup>) as a function of time under MAP conditions, differences in color values were mostly not detected over time or between different storage conditions.

## 274 3.3. Microbiological analysis

Results of the microbiological analysis are presented in Fig. 1. Generally, more rapid growth was 275 observed on all tested media under air when compared to the MAP conditions at the respective 276 storage temperature (4 °C). Initially (day 0), high TPC (Fig. 1A) were typically enumerated on 277 278 MA. The limit of 7.0 log CFU/g was exceeded after 2 days under air storage at 4 °C, whereas at both MAP conditions this limit was reached within 2 days at 8 °C and 4 days at 4 °C. Under low 279 280 O<sub>2</sub> concentrations (L4 and L8), stationary phase was reached after 4 days at 8 °C and 8 days at 4 °C, which closely coincides with the total depletion of oxygen from the package headspace 281 282 (Table 3). TPC of cod packaged under MAP was typically 0.5-1 log higher on MA than on IAL (Fig. 1A-B). 283

LAB enumerated on MRS (Fig. 1C) were able to grow especially well at 8 °C. Oxygen
concentration had little effect on LAB growth under MAs until complete depletion from the
headspace. Respective enumerations were obtained on modified MRS (results not shown). On
the other hand, growth of H<sub>2</sub>S producers (Fig. 1D) was promoted by low oxygen concentrations.
Their growth was highly similar to LAB under low oxygen concentrations (L4 and L8), whereas
stationary growth was observed after 4 days under H8 and six days under H4. Under Air, H<sub>2</sub>S
producers reached higher levels than LAB.

*Pseudomonas* spp. growth (Fig. 1E) was favored by storage under air and effectively inhibited by
elevated carbon dioxide concentrations (60 %). The initial level of *B. thermosphacta* (Fig 1F)
was between 2.5 - 4 log CFU/g and increased by at least 2.5 log CFU/g during storage under all
tested conditions.

## 295 3.4. Quantification of VOCs

296 The VOC concentrations determined by SIFT-MS exceeding the LOQ and having relative

standard deviation below 25 % (Supplementary table 1) are presented in Figures 2-4 as a

function of TPC enumerated on MA. In addition to these compounds, acetone exceeded the LOQ

under L8 (104  $\mu$ g/m<sup>3</sup> on day 7) and ammonia under L4 (9.0  $\mu$ g/m<sup>3</sup> on day 13). When the LOQ

300 was not exceeded, concentration was marked as 0 in Figs 2-4.

301 The differences between the blank averages and the LOQ (Supplementary table 1) could be

attributed to the deviation between blanks. In most cases, concentration of a certain VOC was

303 constant or slightly increasing throughout storage in the blanks. However, ethanol concentration

increased in the MAP blanks by a factor of 1000 or more by the end of storage.

305 *3.4.1. Alcohols* 

306 Levels of alcohols in the package headspace are presented in Fig. 2. Ethanol, 3-methyl-1-307 butanol, isobutyl alcohol and 2,3-butanediol eventually exceeded the LOQ under most of the tested conditions. Ethanol yielded higher concentrations than the other studied compounds. 308 309 However, a high initial ethanol concentration and increasing trend as a function of storage time were typically also detected in the blanks, leading to high LOQs that were only exceeded under 310 311 high O<sub>2</sub> conditions. Concentrations of 3-methyl-1-butanol and 2,3-butanediol started to increase as 7.0 log CFU g<sup>-1</sup> TPC was exceeded and reached up to 500 µg m<sup>-3</sup>. Evolution of these 312 compounds was similar under all tested MAP conditions, whereas lower quantities were 313 314 produced under air. On the other hand, isobutyl alcohol was produced in low quantities and

315 primarily under L4.

316 *3.4.2. Ketones, esters and acids* 

317 During refrigerated storage, two ketones (2-pentanone and acetoin), two esters (ethyl acetate and

ethyl propanoate) and one acid (acetic acid) were analyzed. Due to high relative standard

deviations and/or LOQs, only ethyl acetate, acetic acid and 2-pentanone were quantified (Fig. 3).

320 Increase of ethyl acetate concentration followed a similar trend under every tested condition,

321 whereas other compounds did not exceed the LOQ under all conditions and remained below 200

 $\mu$ g m<sup>-3</sup> throughout storage. Under air storage, only ethyl acetate exceeded the LOQ, whereas

acetic acid was primarily quantified under low O<sub>2</sub> concentrations.

324 *3.4.3. Amine compounds* 

325 Of all tested amine compounds, only trimethylamine (TMA) concentrations increased above the

- LOQ during storage (Fig. 4). At a certain level of microbial growth, higher concentrations of
- 327 TMA were produced under low  $O_2$  concentrations than under high  $O_2$  or air. Under low  $O_2$

328	concentrations at 4 °C (L4), some high concentrations were quantified at relatively low microbial
329	levels. This happened during the late days of storage when TPC was decreasing.
330	3.4.4. Sulfur compounds
331	The time evolution of the sulfur compounds is presented in Fig. 4. Dimethyl disulfide and
332	hydrogen sulfide had a relative standard deviation over 25 %. Relative standard deviation of
333	methyl mercaptan was below 25 % only under condition L8. Dimethyl sulfide (DMS) was
334	typically quantified at low microbial levels (TPC $< 7 \log \text{ CFU g}^{-1}$ ) and was the only sulfuric
335	compound to exceed LOQ under air. Concentrations of DMS did often remain relatively stable
336	throughout storage. Under air or low O2 MAP, higher concentrations were detected than under

high O<sub>2</sub> MAP at a respective level of microbial growth.

#### 338 3.6. Sensory evaluation

339 Figures 2-4 present the individual VOC concentrations as a function of sensory rejection (%). A

sample was considered rejected if labelled as marginal or spoiled. The onset of VOC

341 concentration increase typically coincided with approximately 25 % rejection, irrespective of the

identity of the VOC. At  $\geq$  50 % rejection, TPC enumerated on MA was generally over 7.5 log.

343 Friedman and LSD tests were used for analyzing significant differences among the ranking data.

344 The Friedman test indicated no significant differences ( $\alpha = 0.05$ ) between samples from different

345 days of storage under Air or in the second test of L4, which is why subsequent LSD tests were

not carried out. On the basis of LSD tests (Fig. 5), significant differences ( $\alpha = 0.05$ ) between

samples indicated perceivable change in product quality. Under conditions H4, H8, L4 and L8, a

change in olfactory quality was observed approximately between days 6-8, 3-5, 4-8 and 3-5,

respectively. These changes closely coincide with 50 % rejection (Fig. 2-4).

#### 350 3.7. 16S rRNA gene sequencing

351 Rarefaction curves of samples from intermediate to late days of storage commonly showed trends to level off (Supplementary fig. 1), indicating appropriate sampling depth for most of 352 353 these samples. Even though relatively high species diversity was estimated in samples from early stages of storage (Supplementary table 2), the low read counts were likely insufficient for 354 355 appropriate sampling of diversity. Alpha diversity analysis indicated that ACE-1 (Chao & Lee, 1992) was the only stable richness estimator for the studied dataset (Supplementary table 3). 356 Under modified atmospheres, the ACE-1 index suggested that community richness increased 357 358 during the early days of storage and decreased during the late days, respectively. However, 359 within 95 % confidence intervals, this was only observed under condition H8. On the other hand, diversity indices showed that community diversity was highest in the beginning of storage (day 360 0) under all tested conditions (Supplementary table 4). Diversity was lowest during intermediate 361 storage under MA conditions and in the end of storage under air. 362 After data processing, 503 OTUs were retained at the 97 % sequence identity threshold and a 363

high variation in the number of reads was observed between samples (Supplementary table 2).

365 The relative distribution of the eight most abundant genera is presented in Fig. 6. Initial

366 microbiota (day 0) were generally diverse under all tested conditions. Even though

367 Acinetobacter, Flavobacterium, Photobacterium, Pseudomonas and Psychrobacter were the

368 most abundant genera on day 0, their proportion of the total microbiota was relatively small.

However, under condition L8, *Psychrobacter* and *Flavobacterium* were dominating and a

370 relatively high proportion of *Photobacterium* was detected under H8.

The *Photobacterium* genus became dominant in relative abundance over storage time under all
tested conditions. Initially, *Photobacterium* formed ca. 30 % of the total microbiota under H8

373 and less than 15 % of under H4, L4, L8 and Air. On the later days of storage, over 88 % was detected under both MAP conditions at 4 °C. Under MA conditions, the relative abundance of 374 375 *Photobacterium* was highest during intermediate storage and decreased to some extent by the end of storage, thus increasing community diversity (Supplementary table 4). At higher storage 376 temperature (8 °C), Photobacterium decreased from 78 to 60 % under H8 and 65 to 46 % under 377 L8: under these conditions, Acinetobacter, Brochothrix and Carnobacterium were also able to 378 379 grow. However, a lower number of reads was also obtained from day 7 samples when compared 380 to day 4 samples under these conditions. Under air, 70 and 86 % of Photobacterium was detected 381 on days 2 and 3 of storage.

# 382 **4. Discussion**

383 Growth of SSOs is dependent on the packaging and storage conditions. *Photobacterium* 

*phosphoreum* has been identified as an SSO of marine fish under elevated CO<sub>2</sub> concentrations

385 (Dalgaard et al., 1997; Dalgaard, Mejlholm, & Huss, 1997; Gram & Dalgaard, 2002; Leroi,

2010). In the present study, the *Photobacterium* genus became indeed dominating under all

tested MAP conditions. Since *P. phosphoreum* and *P. iliopiscicarium* are able to grow on MA

388 (Broekaert, Heyndrickx, Herman, Devlieghere, & Vlaemynck, 2011), the results suggest that

389 TPC enumerated on MA reflects the growth of this genus. Furthermore, it was observed that ca.

390 7 log CFU/g is needed for the onset of exponential VOC increase and ca. 7.5 log CFU/g for 50 %

rejection. The results are in line with Dalgaard et al. (1997) for cod fillets stored under 60/40/0

and 60/0/40 (CO<sub>2</sub>/O<sub>2</sub>/N<sub>2</sub>) at 0 °C. The results thus suggest that representatives of the

393 *Photobacterium* genus contribute to the increase in VOC concentrations and sensory rejection

and that the onset of exponential VOC increase can be observed at relatively high microbial

395 levels.

396	When stored under air, Pseudomonas and Shewanella spp. have commonly been considered as
397	SSOs of refrigerated or iced marine fish (Gram & Huss, 1996; Gram & Dalgaard, 2002).
398	Parlapani et al. (2015) observed <i>Pseudomonas</i> and H <sub>2</sub> S producers to be dominating in sea bass
399	stored both under air and MAP (60/10/30 % CO <sub>2</sub> /O <sub>2</sub> /N <sub>2</sub> ); under MAP, LAB and B.
400	thermosphacta were observed to be co-dominating. Respectively, both enumeration (Fig. 1) and
401	sequencing (Fig. 6) results of the present study indicate Pseudomonas growth under air and
402	inhibition under MAP. Under MA conditions, high carbon dioxide concentrations are known to
403	inhibit pseudomonads (Gram & Huss, 1996).
404	Enumeration of <i>Photobacterium</i> can be affected by the properties of the growth media. Broekaert
405	et al. (2011) identified MA to be more suitable for the enumeration of marine bacteria than IAL.
406	In the present study, the difference typically observed between TPC enumerated on MA and IAL
407	likely reflects the dominance of the Photobacterium genus. In the beginning of storage (day 0),
408	highly similar results are obtained on both media, whereas higher counts are generally
409	enumerated on MA on later days of storage (Fig. 1). According to the oligotyping results (Fig.
410	6), respectively, the <i>Photobacterium</i> genus typically forms a small fraction of the initial
411	microbiota and majority at later stages of storage. Since P. phosphoreum is sensitive to heat,
412	pour plating temperatures (< 50 °C) have been suggested to lead into underestimation of its
413	growth (Dalgaard et al., 1997). Incubation temperature of 23-25 °C or higher has also been
414	suggested to inhibit P. phosphoreum growth (Dalgaard et al., 1997); however, similar
415	enumeration results were obtained in the present study on MA incubated at 22 or 15 °C (results
416	not shown).

417 Elevated CO<sub>2</sub> concentrations have been reported to favor the growth of CO<sub>2</sub> tolerant LAB (Gram & Dalgaard, 2002; Leroi, 2010). Analogously, the present enumeration results on MRS suggests 418

419 that facultative anaerobic LAB were able to grow under both MAP conditions. Even though 420 acetate-containing MRS has been reported to inhibit certain LAB such as carnobacteria (Leroi, 2010), comparative enumeration on MRS and mMRS resulted in highly similar CFU levels 421 422 (results not shown). According to the sequencing results (Fig. 6), the relative abundance of 423 carnobacteria was higher under MA conditions when compared to storage under air. High CO<sub>2</sub> 424 concentration also had an inhibitive effect on H<sub>2</sub>S producers under MAP when compared to air storage, especially under high O<sub>2</sub> conditions. An additional inhibitive effect of O<sub>2</sub> was also 425 observed by López-Caballero et al. (2001), which was suggested to be due to synergistic effect 426 427 between the gases.

428 An increase in concentrations of several alcohols was detected in the present study. Respectively, 429 ethanol, 3-methyl-1-butanol and 2,3-butanediol have frequently been identified as potential spoilage indicators of marine fish under air and/or MAP in several studies (Duflos, Coin, Cornu, 430 431 Antinelli, & Malle, 2006; Mikš-Krajnik et al., 2016; Olafsdottir et al., 2005; Parlapani et al., 432 2014; Parlapani et al., 2015; Parlapani et al., 2015). Olafsdottir et al. (2005) observed increasing concentrations of ethanol, 2-methyl-1-propanol, 3-methyl-1-butanol and 2,3-butanediol for 433 aerobically stored cod fillets at 0.5 °C. Ethanol and 2-methyl-1-propanol were suggested to have 434 435 importance in early detection of spoilage despite non-continuous increase. Duflos et al. (2006) found several alcohols including ethanol, 3-methyl-1-butanol and 2,3-butanediol to increase in 436 437 cod, mackerel and whiting stored under vacuum at 4 °C for ten days. Production of different 438 alcohols has been associated with several microbial species among LAB, *Shewanella*, 439 Pseudomonas, P. phosphoreum and B. thermosphacta (Casaburi, Piombino, Nychas, Villani, & 440 Ercolini, 2015; Hernández-Macedo et al., 2012; Noseda et al., 2012). In the present study, the

441 dominance of *Photobacterium* suggests that the production of alcohols could be largely442 attributed to this genus.

The H<sub>2</sub>S concentrations remained low under all tested conditions. Low H<sub>2</sub>S production by *P*. *phosphoreum* has been observed in other studies (Dalgaard, Gram, & Huss, 1993). In Danish
marine fish, *Shewanella baltica* has been identified as the main H<sub>2</sub>S producer (Vogel et al.,
2005). Even though *S. putrefaciens* has high spoilage potential due to the production of intensive
off-odors, high levels (8 log CFU/g) are needed for off-odor production (Dalgaard, 1995). The
present results thus support the conclusion that significant VOC production can only be observed
at relatively high microbial levels.

450 TMA is produced by bacteria that utilize trimethylamine oxide (TMAO) for anaerobic respiration

451 and results in ammonia-like or "fishy" odors characteristic for spoiled marine fish (Gram &

452 Dalgaard, 2002). Oxygen has been observed to inhibit the reduction of TMAO into TMA as well

as to reduce the growth of TMA-producing *P. phosphoreum* (Boskou & Debevere, 1997;

454 Dalgaard et al., 1997). This is in line with the results of the present study. Since TMA

455 concentration was notably higher at 50 % rejection than its human olfactory threshold (OT)  $6 \mu g$ 

m<sup>3</sup> (Devos, Patte, Rouault, Laffort, & Van Gemert, 1990), TMA was likely to contribute to the
rejection of the samples.

Even though VOCs are often produced in low quantities, their effect on the perceived quality of

the fish can be significant if they have low OTs. Alcohols have generally high OTs, whereas

460 sulfur and amine compounds often become detectable at very low quantities (Devos et al., 1990).

461 However, OTs are commonly determined for single compounds from a continuous airflow.

462 Furthermore, OT values indicate the lowest quantity of a VOC that can be perceived by the

panelists, instead of indicating whether it is considered acceptable. Acceptance of an odor may
depend on cultural, social and economic aspects, as well as the characteristics of the food
product. Since olfactory evaluation of fish freshness is based on the overall smell, OTs and
acceptability of VOCs are likely dependent on the composition of the whole VOC profile.
Instead of using single compounds for quality and spoilage evaluation of fish, multiplecompound quality indices have shown promising potential (Jørgensen et al., 2001).

In the present study, the concentration of several VOCs increased as a function of microbial 469 growth. Under most of the tested conditions, increase in 2.3-butanediol, ethanol, ethyl acetate, 3-470 methyl-1-butanol and trimethylamine were observed. All these compounds have been recognized 471 472 as fish spoilage metabolites (Duflos et al., 2006; Olafsdóttir et al., 1997). For example, ethanol, 473 ethyl acetate and/or 3-methyl-1-butanol have also been associated with the spoilage of several non-seafood products packaged under modified atmospheres (Casaburi et al., 2015; Nieminen, 474 475 Dalgaard, & Björkroth, 2016; Zhang et al., 2013; Zhang, Samapundo, Pothakos, Sürengil et al., 476 2013): monitoring of such compounds could enhance the applicability of an intelligent packaging solution into a wider range of food products. Since the *Photobacterium* genus was highly 477 478 abundant under all storage conditions, differences in its metabolism could contribute to the observed differences in the VOC profiles between the tested storage conditions. Onset of 479 exponential concentration increase was typically observed between TPC 7 - 7.5 log CFU/g and 480 25 - 50 % rejection. Respectively, late increase of VOC concentrations in relation to microbial 481 growth has also been detected in other studies (Olafsdottir et al., 2005). Detection of early 482 483 spoilage thus requires that low concentrations of relevant VOCs can be detected.

#### 484 **5.** Conclusions

Different packaging and storage conditions affect the evolution of fish microbiota and the generated VOCs in the package headspace. In the present study, the SIFT-MS technology allowed the real-time quantification of VOCs directly from the package headspace. This approach eliminated the need of sample preparation procedures, while allowing fast and sensitive analysis of the VOC profile over storage time. The obtained results directly represent the quality deterioration of fish and thus the reality that a sensor needs to be able to respond to during storage.

Packaging and storage conditions affect the evolution of the VOC profile and should be
considered in the selection of spoilage indicators. In the present study, increase in 2,3-butanediol,
ethanol, ethyl acetate, 3-methyl-1-butanol and trimethylamine concentrations during storage
suggests that these compounds could be used in detecting spoilage of raw Atlantic cod. However,
since VOC concentrations typically remain at low quantities even at the late stage of storage,
detection of early spoilage calls for sensitivity for low concentration ranges.

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- 788 Figure captions
- **Fig. 1.** Counts of total viable psychrotrophic bacteria (A-B), lactic acid bacteria (C), H<sub>2</sub>S
- 790 producers (D), pseudomonads (E) and *Brochothrix thermosphacta* (F) in Atlantic cod fillet
- 791 portions stored under conditions H4 (60 % CO<sub>2</sub> /40 & O<sub>2</sub>/0 % N<sub>2</sub> at 4 °C), H8 (60/40/0 8 °C), L4
- 792 (60/5/35 4 °C), L8 (60/5/35 8 °C) and air (4 °C).
- **Fig. 2.** Concentrations ( $\mu$ g m<sup>-3</sup>) of alcohols quantified by SIFT-MS as a function of total viable
- psychrotrophic counts (TPC) or sensory rejection % in Atlantic cod fillet portions stored under
- 795 conditions H4 (60 % CO<sub>2</sub> /40 & O<sub>2</sub>/0 % N<sub>2</sub> at 4 °C), H8 (60/40/0 8 °C), L4 (60/5/35 4 °C), L8
- 796 (60/5/35 8 °C) and air (4 °C).
- **Fig. 3.** Concentrations (µg m<sup>-3</sup>) of ketones, esters and acids quantified by SIFT-MS as a function
- of total viable psychrotrophic counts (TPC) or sensory rejection % in Atlantic cod fillet portions
- 799 stored under conditions H4 (60 % CO<sub>2</sub> /40 & O<sub>2</sub>/0 % N<sub>2</sub> at 4 °C), H8 (60/40/0 8 °C), L4 (60/5/35
- 800 4 °C), L8 (60/5/35 8 °C) and air (4 °C).
- **Fig. 4.** Concentrations ( $\mu$ g m<sup>-3</sup>) of amines and sulfur compounds quantified by SIFT-MS as a
- 802 function of total viable psychrotrophic counts (TPC) or sensory rejection % in Atlantic cod fillet
- 803 portions stored under conditions H4 (60 % CO<sub>2</sub> /40 & O<sub>2</sub>/0 % N<sub>2</sub> at 4 °C), H8 (60/40/0 8 °C), L4
- 804 (60/5/35 4 °C), L8 (60/5/35 8 °C) and air (4 °C).
- **Fig. 5.** Ranks (1=least fresh, 4=most fresh) assigned to cod fillet samples from four different
- 806 days of storage under conditions H4 (60 % CO<sub>2</sub> /40 & O<sub>2</sub>/0 % N<sub>2</sub> at 4 °C), H8 (60/40/0 8 °C), L4
- 807 (60/5/35 4 °C) and L8 (60/5/35 8 °C). Storage days with different postscripts (a-c) within a
- so condition are significantly different (p < 0.05).

- **Fig. 6.** Composition of microbiota in Atlantic cod fillet portions stored under conditions H4 (60
- 810 % CO<sub>2</sub> /40 & O<sub>2</sub>/0 % N<sub>2</sub> at 4 °C), H8 (60/40/0 8 °C), L4 (60/5/35 4 °C), L8 (60/5/35 8 °C) and
- air (4 °C), determined by amplicon (NGS) sequencing of the 16S rRNA gene.

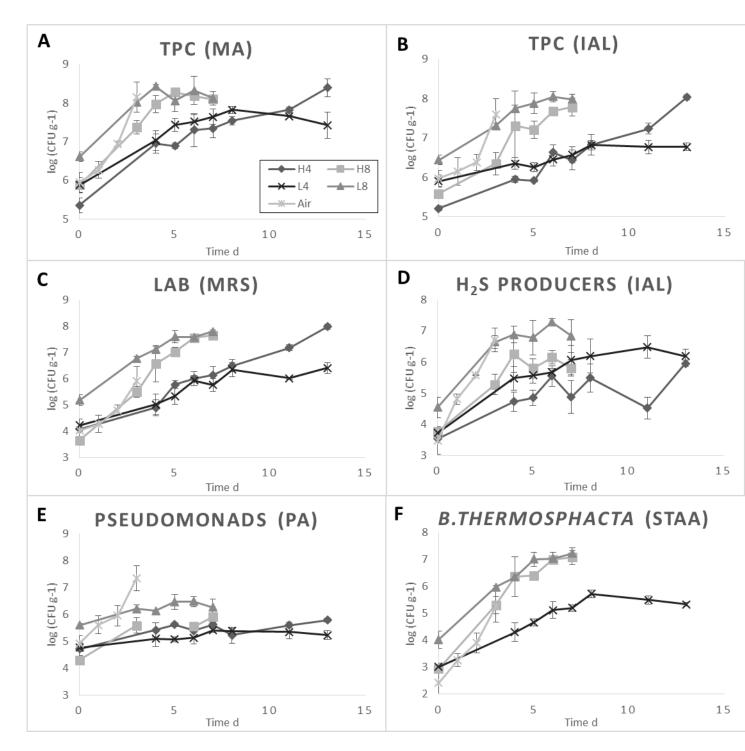
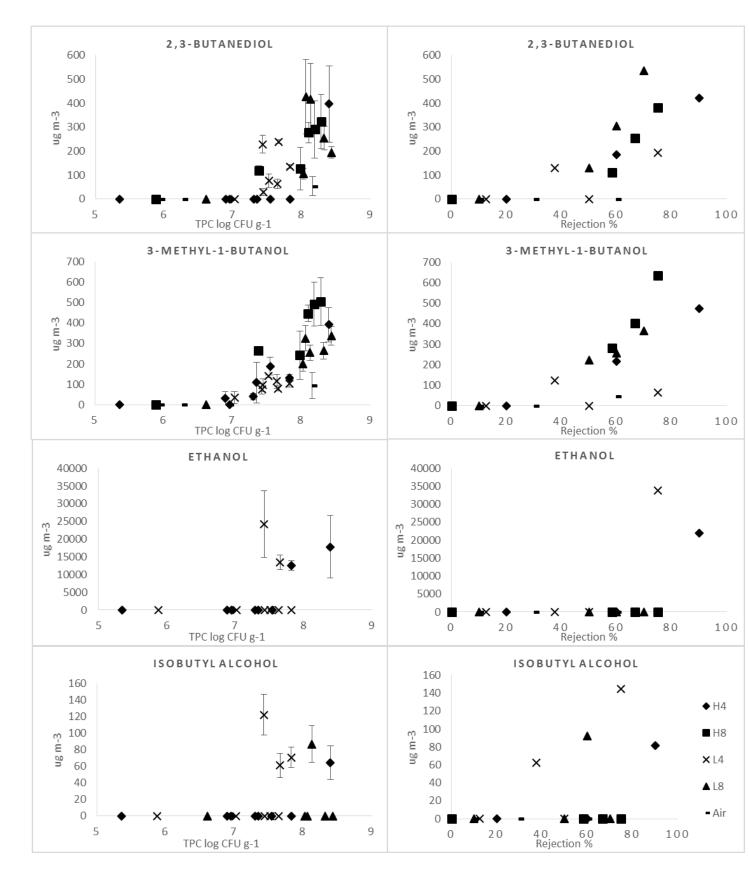


Fig. 1.





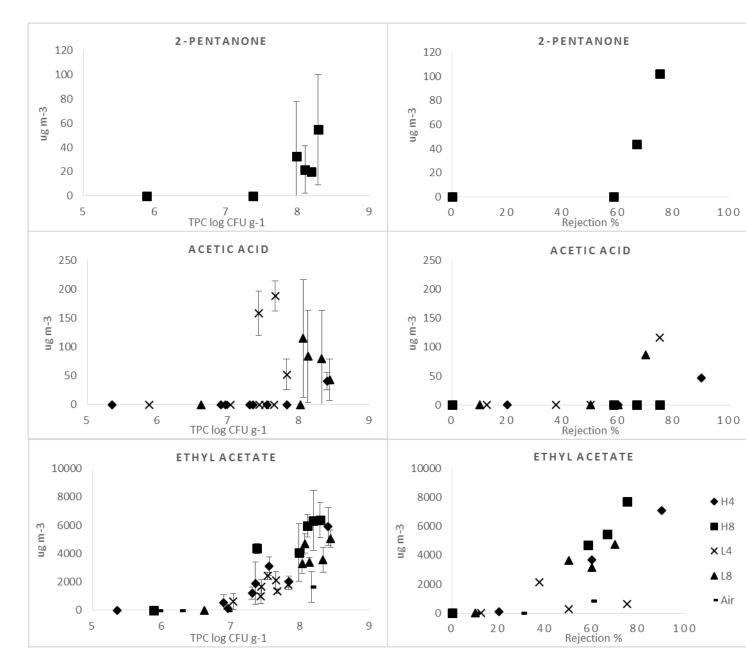
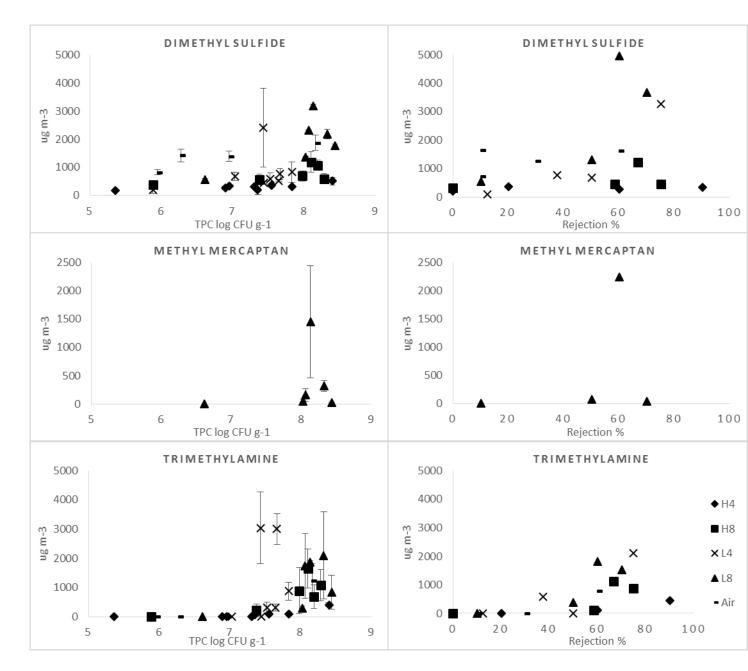


Fig. 3.





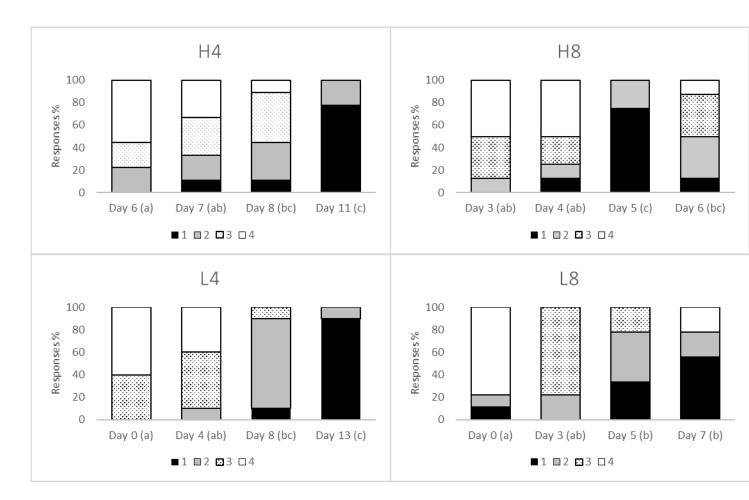


Fig. 5.

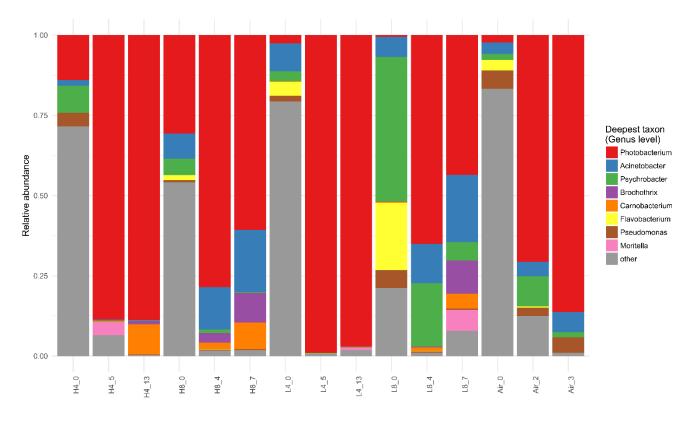




 Table 1. Packaging and storage conditions used in the study. Samples from days denoted with (\*)

 were studied by amplicon sequencing.

	H4	H8	L4	L8	Air
Headspace gases	60/40/0	60/40/0	60/5/35	60/5/35	air
(% CO <sub>2</sub> /O <sub>2</sub> /N <sub>2</sub> )					
Temperature	4	8	4	8	4
(°C)					
Days of analysis	0*,4,5*,6,7,8,11,13*	0*,3,4*,5,6,7*	0*,4,5*,6,7,11,13*	0*,3,4*,5,6,7*	0*,1,2*,3*

VOC	Precursor ion	m/z	b (%)	k	Product ion
Acids					
Acetic acid	$H_3O^+$	61	100	2.6 E -09	$CH_3COOH_2^+$
	$NO^+$	90	100	9.0 E -10	NO <sup>+</sup> .CH <sub>3</sub> COOH
	$O_2^+$	60	50	2.3 E -09	CH <sub>3</sub> COOH <sup>+</sup>
Alcohols					
Ethanol	$H_3O^+$	47	100	2.7 E -09	$C_2H_7O^+$
	$H_3O^+$	65			$C_2H_7O^+.H_2O$
	$H_3O^+$	83			$C_2H_7O^+.(H_2O)_2$
2,3-butanediol	$\mathbf{NO}^+$	89	100	2.3 E -09	$C_4H_9O_2^+$
3-methyl-1-butanol	$H_3O^+$	71	100	2.8 E -09	$C_{5}H_{11}^{+}$
	$NO^+$	87	85	2.3 E -09	$C_5H_{11}O^+$
isobutyl alcohol	$NO^+$	73	95	2.4 E -09	$C_4H_9O^+$
-	$O_2^+$	33	50	2.5 E -09	$CH_5O^+$
Aldehydes					
2-methylpropanal	$O_2^+$	72	70	3.0 E -09	$C_4H_8O^+$
3-methylbutanal	$NO^+$	85	100	2.4 E -09	$C_5H_9O^+$
Ketones					
Acetone	$H_3O^+$	59	100	3.9 E -09	$C_3H_7O+$
	$NO^+$	88	100	1.2 E -09	$NO^+.C_3H_6O$
Acetoin	$O_2^+$	88	20	2.5 E -09	$C_4H_8O_2^+$
2-pentanone	$NO^+$	116	100	3.1 E -09	$NO^{+}.C_{5}H_{10}O^{+}$
Sulfur compounds					
Hydrogen sulfide	$H_3O^+$	35	100	1.6 E -09	$H_3S^+$
	$O_2^+$	34	100	1.4 E -09	$H_2S^+$
Methyl mercaptan	$H_3O^+$	49	100	1.8 E -09	$CH_4S.H^+$
Dimethyl sulfide	$H_3O^+$	63	100	2.5 E -09	$(CH_3)_2S.H^+$
	$NO^+$	62	100	2.2 E -09	$(CH_3)_2S^+$
Dimethyl disulfide	$H_3O^+$	95	100	2.6 E -09	$(CH_3)_2S_2.H^+$
J.	$NO^+$	94	100	2.4 E -09	$(CH_3)_2S_2^+$
	$O_2^+$	94	80	2.3 E -09	$(CH_3)_2S_2^+$
Dimethyl trisulfide	$H_3O^+$	127	100	2.8 E -09	$C_2H_6S_3H^+$
•	$NO^+$	126		1.9 E -09	$C_2H_6S_3^+$
Esters					
Ethyl acetate	$NO^+$	118	90	2.1 E -09	NO <sup>+</sup> .CH <sub>3</sub> COOC <sub>2</sub> H <sub>5</sub>
Ethyl propanoate	$H_3O^+$	103	95	2.9 E -09	$C_2H_5COOC_2H_5.H^+$
	NO <sup>+</sup>	132	60	2.5 E -09	NO <sup>+</sup> .C <sub>2</sub> H <sub>5</sub> COOC <sub>2</sub> H <sub>5</sub>
Amines					
Ammonia	$H_3O^+$	18	100	2.6 E -09	$\mathrm{NH_{4^+}}$
	$O_2^+$	17	100	2.4 E -09	NH <sub>3</sub> <sup>+</sup>
Dimethylamine	$H_3O^+$	46	100	2.1 E -09	$(CH_3)_2N.H^+$
Trimethylamine	$H_3O^+$	60	90	2.0 E -09	$(CH_3)_3N.H^+$
,	NO <sup>+</sup>	59	100	1.6 E -09	$(CH_3)_3N^+$

**Table 2.** Product ions of volatile organic compounds (VOCs) quantified with SIFT-MS, respectivemass to charge ratios (m/z), branching ratios (b) and reaction rate coefficients (k).

*Table 3.* Headspace gases ( $O_2$ ,  $CO_2$ ), pH and color ( $L^*$ ,  $a^*$ ,  $b^*$ ) as a function of time under conditions H4 (60 %  $CO_2$  /40 &  $O_2$ /0 %  $N_2$  at 4 °C),

	Series	Time of storage (d)										
		0	1	2	3	4	5	6	7	8	11	13
02	H4	$41.17\pm0.15$				$51.9 \pm 1.6$	$51.77 \pm 2.12$	$49.8 \pm 1.9$	$49.33 \pm 1.9$	$48.5\pm1.48$	$43.97 \pm 1.9$	$37.77 \pm 2.98$
	H8	$42.47\pm0.81$			$49.63 \pm 1.27$	$39.57 \pm 18.8$	$47.57\pm0.58$	$46.17 \pm 1.53$	$46.7\pm0.78$			
	L4	$5.09\pm0.12$				$4.26\pm0.18$	$2.87\pm0.44$	$2.41\pm0.79$	$1.67\pm0.52$	$0.25\pm0.3$	$0.01 \pm 0$	$0\pm 0$
	L8	$4.47\pm0.51$			$4.36\pm0.09$	$2.81\pm0.87$	$0.73\pm0.16$	$0.07\pm0.11$	$0\pm0.01$			
	Air	$20.73\pm0.06$	$20.3\pm0.1$	$19.67\pm0.06$	$16.53 \pm 1.4$							
CO2	H4	$55.83 \pm 0.25$				$44.07\pm0.65$	$43.43 \pm 1.97$	$43.47\pm0.83$	$46.17 \pm 1.57$	$47\pm1.9$	$52.27 \pm 1.45$	$53.7 \pm 12.38$
	H8	$55.43 \pm 2.12$			$47 \pm 1.04$	$37.5 \pm 13.6$	$49.33\pm0.38$	$50.17 \pm 1.86$	$49.77\pm0.45$			
	L4	$56.27 \pm 1.02$				$42.87\pm0.51$	$41.1\pm2.17$	$44.17\pm2.27$	$41.07\pm0.55$	$42.47\pm2.08$	$40.53 \pm 1.79$	$44.83 \pm 4.47$
	L8	$56.47 \pm 0.51$			$45.13\pm1$	$43.77\pm2.4$	$47.87 \pm 4.11$	$45.53 \pm 1.61$	$48.47 \pm 2.93$			
	Air	$0.27\pm0.06$	$1.37\pm0.21$	$1.3\pm0$	$4.77 \pm 1.31$							
рН	H4	$6.34\pm0.16$				$6.2\pm0.11$	$6.36\pm0.13$	$6.34\pm0.15$	$6.33\pm0.09$	$6.41\pm0.08$	$6.23\pm0.04$	$6.46\pm0.08$
	H8	$6.42\pm0.04$			$6.41\pm0.13$	$6.66 \pm 0.21$	$6.6\pm0.01$	$6.54\pm0.1$	$6.78\pm0.04$			
	L4	$6.68\pm0.24$				$6.3\pm0.1$	$6.4\pm0.08$	$6.38\pm0.04$	$6.62\pm0.14$	$6.59\pm0.06$	$6.67\pm0.09$	$6.57\pm0.06$
	L8	$6.88 \pm 0.07$			$6.56\pm0.04$	$6.71\pm0.02$	$6.48\pm0.17$	$6.71 \pm 0.1$	$6.75\pm0.06$			
	Air	$6.68\pm0.07$	$6.72\pm0.02$	$6.68\pm0.08$	$6.7\pm0.05$							
L*	H4	$59.08 \pm 1.77$				$62.36 \pm 5.41$	$60.3 \pm 1.59$	$58.01 \pm 2.22$	$64.15\pm2.09$	$61.2\pm2.57$	$66.12\pm2.06$	$61.03 \pm 2.3$
	H8	$58.15 \pm 2.29$			$58.37 \pm 1.15$	$60.63 \pm 1.32$	$60.49 \pm 0.41$	$56.9\pm2.06$	$58.63 \pm 0.82$			
	L4	$56.33 \pm 1.68$				$61.1 \pm 1.81$	$59.94 \pm 2.66$	$59.94 \pm 1.25$	$59.32 \pm 1.09$	$60.97 \pm 1.16$	$60.63 \pm 0.69$	$62.53 \pm 1.3$
	L8	$61.8\pm2.37$			$59.94 \pm 2.47$	$59.97 \pm 1.7$	$62.13 \pm 1.64$	$62.84\pm0.35$	$60.86 \pm 0.62$			
	Air	$55.7 \pm 1.06$	$58.71 \pm 1.33$	$58.71 \pm 1.54$	$56.71 \pm 1.46$							
a*	H4	$\textbf{-2.5} \pm 0.18$				$\textbf{-2.7} \pm 0.07$	$\textbf{-2.85} \pm 0.31$	$-2.61\pm0.09$	$-3 \pm 0.05$	$-2.91\pm0.28$	$\textbf{-2.97} \pm 0.16$	$-2.63\pm0.3$
	H8	$\textbf{-2.43} \pm 0.11$			$\textbf{-2.78} \pm 0.28$	$\textbf{-2.88} \pm 0.05$	$\textbf{-3.09} \pm 0.21$	$-3.14\pm0.2$	$\textbf{-2.99} \pm 0.28$			
	L4	$\textbf{-2.6} \pm 0.45$				$-3.11\pm0.12$	$-2.64\pm0.23$	$-2.95\pm0.14$	$-3.05\pm0.3$	$-3.23\pm0.46$	$\textbf{-3.06} \pm 0.06$	$-3.41\pm0.3$
	L8	$\textbf{-3.07} \pm 0.37$			$-3.21\pm0.48$	$-3.07\pm0.44$	$\textbf{-3.59} \pm 0.16$	$-3.35\pm0.23$	$-3.55\pm0.12$			
	Air	$\textbf{-2.55} \pm 0.33$	$-2.65\pm0.16$	$-2.5\pm0.17$	$\textbf{-2.98} \pm 0.07$							
<i>b</i> *	H4	$-1.64 \pm 1.45$				$0.28 \pm 1.05$	$\textbf{-1.03} \pm 0.16$	$\textbf{-2.41} \pm 1.11$	$\textbf{-0.03} \pm 0.96$	$0.49\pm0.53$	$1.66 \pm 2.9$	$2.68 \pm 1.88$
	H8	$\textbf{-1.09} \pm 1.78$			$0.97 \pm 1.72$	$\textbf{-0.99} \pm 1.14$	$0.61 \pm 1.42$	$0.64\pm0.16$	$\textbf{-1.27} \pm 1.52$			
	L4	$\textbf{-1.15} \pm 0.88$				$\textbf{-0.06} \pm 0.98$	$\textbf{-0.14} \pm 1.86$	$\textbf{-1.21} \pm 0.61$	$0.54 \pm 2.32$	$0.05 \pm 1.36$	$\textbf{-0.06} \pm 1.96$	$\textbf{-0.42} \pm 1.03$
	L8	$-2.97\pm0.37$			$0.48 \pm 1.13$	$0.13 \pm 0.59$	$\textbf{-0.32} \pm 1.8$	$0.01\pm0.65$	$-1 \pm 0.88$			
	Air	$-2.97 \pm 1.29$	$-1.73\pm2.38$	$\textbf{-3.2}\pm0.73$	$-2.06\pm0.52$							

H8 (60/40/0 8 °C), L4 (60/5/35 4 °C), L8 (60/5/35 8 °C) and Air (air 4 °C).